

The Studying of Declining Reservoir Pressure on Natural Gas Sweetening Process: A Case Study and Simulation

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Abstract

Natural gas is considered as one of the most popular source of energy in recent era and future as well. However, raw natural gas usually contents several of nonhydrocarbon components for instance, hydrogen sulphide and carbon dioxide. Indeed, these impurities are undesirable compounds and cause many problems for example, corrosion and environment pollution. Moreover, amine gas sweetening process is considered the most popular technology to remove acid gases from natural gas stream. However, when the hydrocarbons reservoir pressure declines, new wells are drilled to maintain production and enhanced oil recovery methods are also applied at the end age of the reservoir. As a result, the declining of the reservoir pressure leads to decline the operation pressure for the amine contactor tower and it may lead to significant effects on gas sweetening process efficiency and performance. Therefore, this study aims to simulate gas-sweetening process for given raw natural gas stream that it contents high quantity of acid gases by using Aspen HYSYS simulator program and then examine the effects of declining reservoir pressure on the sweetening process. The case study gas stream operation pressure is about 7000 Kpa. However, the study found that when the sour gas pressure declined that will effect acid gases loading in amine solution.

Index terms— Natural gas sweetening, reservoir pressure, amine solution, process simulation, Aspen Hysys, process efficiency.

1 Introduction

he demand of natural gas in recent decade has been dramatic. In fact, natural gas poses a huge rule in the recent world economy and development. However, natural gas usually contains several impurities for instance, acid gases that it need to be removed from natural gas to meet the gas pipelines specifications. Stewart and Arnold (2011) note that gas contracts restrict H₂S content about 4ppm and CO₂ about 2% in natural gas stream. Thus, several gas sweetening are develop in order to remove acid gases Author ? : School of chemical and petroleum engineering, Koya University, Kurdistan region-Iraq.

Author ? : School of Science & Engineering, Teesside university, UK. from raw natural gas stream for example, chemical absorption, solid bet sweetening method and physical absorption method. However, amine gas sweetening may consider the most common process among natural gas sweetening method. Indeed, amine gas sweetening process has many advantages for instance, continues process, the ability to regenerate the process solvent. However, any amine process has several operation conditions for instance amine contactor pressure and amine solution concentration. In fact, amine contactor pressure is considered one of the most important amine process operation condition. Moreover, any declining of absorber pressure may affect the whole process. Therefore, a case study for raw natural gas stream will be examined by Aspen HYSYS simulation and the operation pressure of amine absorber tower will be examined by changing the value of amine contactor pressure several time and then transfer the results to MS excel to find out the results of effect of declining reservoir

5 RESULT AND DISCUSSION

43 pressure on amine process. Moreover, the given gas stream may consider quite sour gas because is contents a
44 high quantities of H₂S about (5.3%) and CO₂ about (4.4%). However, amine gas sweetening process will be
45 able to reduce acid gases contents by using a suitable amine solvent type.

2 II.

3 Amine Process Description

48 This method is also included several processes that utilized different chemical solvents:?
49 MEA(MonoEthanolAmine) process ? DEA(DiEthanolAmine) process ? MDEA(MethetylDiEthanolAmine)
50 process ? DGA(DiGlycolAmine) process ? Hot potassium carbonate process

51 The chemical reaction of amines with H₂S and CO₂ Could be summarized below: $2RNH_2 + H_2S = (RNH_3)2S$
52 $2RNH_2 + CO_2 = RNHCOONH_3R$ *R= mono,di,tri-Ethanol Chemical solvent method may consider the most
53 remarkable and successful method in Natural gas sweetening field. Moreover, it may consider number one in most
54 gas plant around world. Indeed, this method is utilized an aqueous solution of a weak base to chemically react
55 with and absorb the acid gases in the natural gas stream (Stewart and Arnold, 2011). In fact, these chemical
56 solvents possess high affinity toward acid gases. The aqueous solution could be regenerating easily and recirculate
57 to the process. The chemical solvent method is mainly utilized either Amine or carbonate solution to achieve
58 the sweetening process. The amine gas sweetening process is considered a continues process and figure (1) shows
59 the general flow diagram for amine gas sweetening plant. Firstly, sour gas stream is usually enters to scrubber
60 to remove sour gas constants. Secondly, sour gas enters to the bottom side of amine absorber tower and flow
61 countercurrent to amine solvent and Sweet gas will leave the top of the contactor tower and need to be processed
62 to dehydration process to remove saturated water. Moreover, Dirty or rich amine will leave bottom of contactor
63 tower and need to be regenerate. Finally, Amine stripping tower (regenerator) is used to regenerate the dirty
64 amine hot lean amine need to be cooled therefore it flows to amine heat exchanger and then back to contactor
65 tower. The brief of amine process could be described as following:

4 Case Study

67 The case study gas composition is shown in table (1). It seems that the gas has high content of acid gases.
68 However, the gas analyzed on dry basis. Therefore, gas water content should be calculated. After achieving
69 above, the simulation environment is entered. simulation environment may consider the main simulation area,
70 which it deals with the plant and shows the FPD for the process. It quite important to uses inlet gas separator
71 to remove any undesirable impurities such as, solid particulars and liquids. Amine contactor is also important
72 part from the plant which it also need some specifications for example, streams temperature and pressure and
73 the amine (DEA) concentration (35% by wt. is used) and figure (3) shows amine contactor menu. After finishing
74 above steps amine heat exchanger is also installed. Moreover, dirty amine needs to be regenerate and that could
75 be achieved by installing the amine regenerator after amine heat exchanger and figure (4) shows amine regenerator
76 menu. The simulation process done successfully and figure (5) shows process flow diagram. As it seems from
77 figure, (5) several processes unite are used in amine process. Infect, installing flash separator for rich amine is
78 quite important in order to avoid any technical problems. Moreover, the ADJUST function is also important to
79 adjust the mass flow rate of lean amine with the H₂S molar friction in sweet gas stream. In addition, water make
80 up stream should be added with a mixer to the process. In fact, amine concentration may be built up in the
81 process because of water and amine losses with sweet gas. Therefore, water makes up stream will maintain and
82 support the concentration of DEA at acceptable value. The simulation process done and the process achieved
83 high acid gas removal that it will be discussed in result and discussion part. b) Effects of declining pressure on
84 gas sweetening process At the starting age of reservoir production, the reservoir pressure is usually sufficient to
85 produce oil and gas as well. Moreover, it drives by natural reservoir mechanisms for example, water. Moreover,
86 the wellhead pressure could be decreed or controlled by wellhead choke. As it showed that from given, gas stream
87 data the reservoir peruse is high about 7000 Kpa. However, after period of reservoir production time the pressure
88 will gradually declines. Thus, in this study part several operation pressure will be examined for example, 7000
89 Kpa, 6000Kpa,5000Kpa and 2000 Kpa. These values will be applied in previous HYSYS simulation program and
90 the effects pressure change can be recognized for each case. Indeed, the declining of absorber pressure may lead
91 to decrees the capacity of contactor unit.

5 Result And Discussion

93 Rich amine loading results for several assumed operation pressures are recognized and the whole results are
94 transferred to MS excel program and the results can be showed in figure (7): It seems that from figure (7), the
95 decreasing of amine contactor tower pressure due to declining of reservoir pressure has a direct effect on the rich
96 amine lodging. As a result, the rich amine loading (moles of Acid gases/moles of amine) will be reduced and
97 that will significantly effects the sweetening process performance. Moreover, in this case amine circulation rate
98 should be increased in order to meet the product requirements and this will effect directly on process operation
99 cost. Thus, drilling new wells or using EOR methods are the most economical and successful 4. Maddox, R.
100 (1982) Gas conditioning and processing: gas and liquid sweetening. Oklahoma: Campbell petroleum.

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102 actions to overcome this problem. Thus, it can argue that if pressure of the amine contactor unite decreases
103 due to decline in reservoir pressure then the partial pressure of Acid gases such as CO 2 will also be reduced.
104 V.

105 6 Conclusion

106 In conclusion, this study is attempted to examine the effect of declining the reservoir pressure on gas sweetening
107 process. Moreover, it is also simulated the gas sweetening process by Aspen HYSYS program. It can argue that
108 the declining of reservoir pressure is directly affected amine gas sweetening process efficiency and performance
109 and several technical problems for instance, it lead to increase the lean amine circulation rate, decrees rich amine
110 acid gases lodging, increase in energy consumption by process and increase the operation cost. Moreover, it
111 strongly recommended that incrusting the amine type or concentration in the process. However, this only could
112 apply for short time because high amine concentration means high corrosion and cost. Therefore, the maintain
113 of reservoir pressure by drilling additional well and using enhanced oil recovery method may consider the best
solution to solve this problem.



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Figure 1: Figure 1 :

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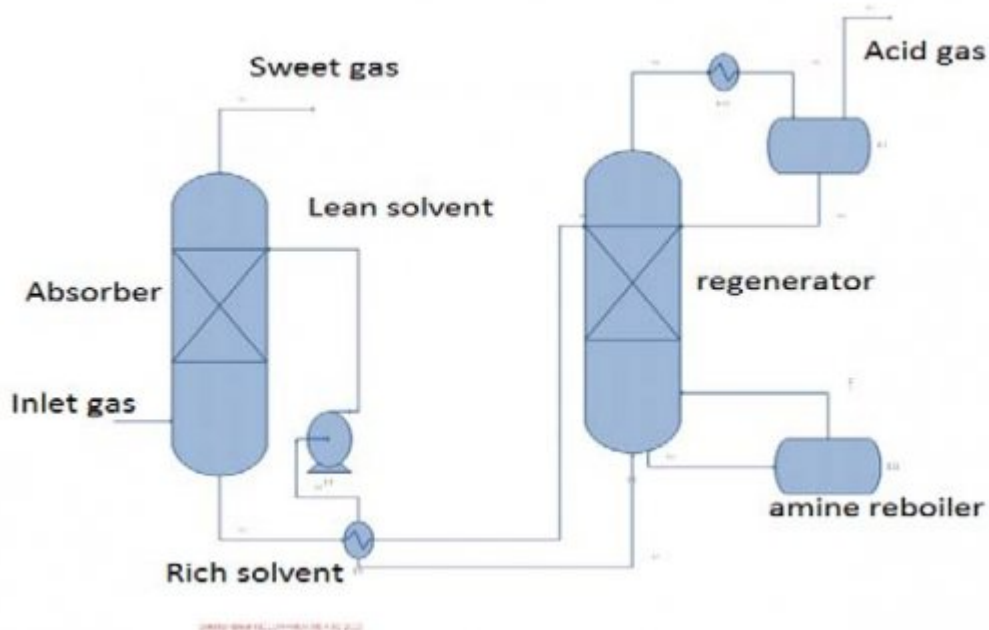
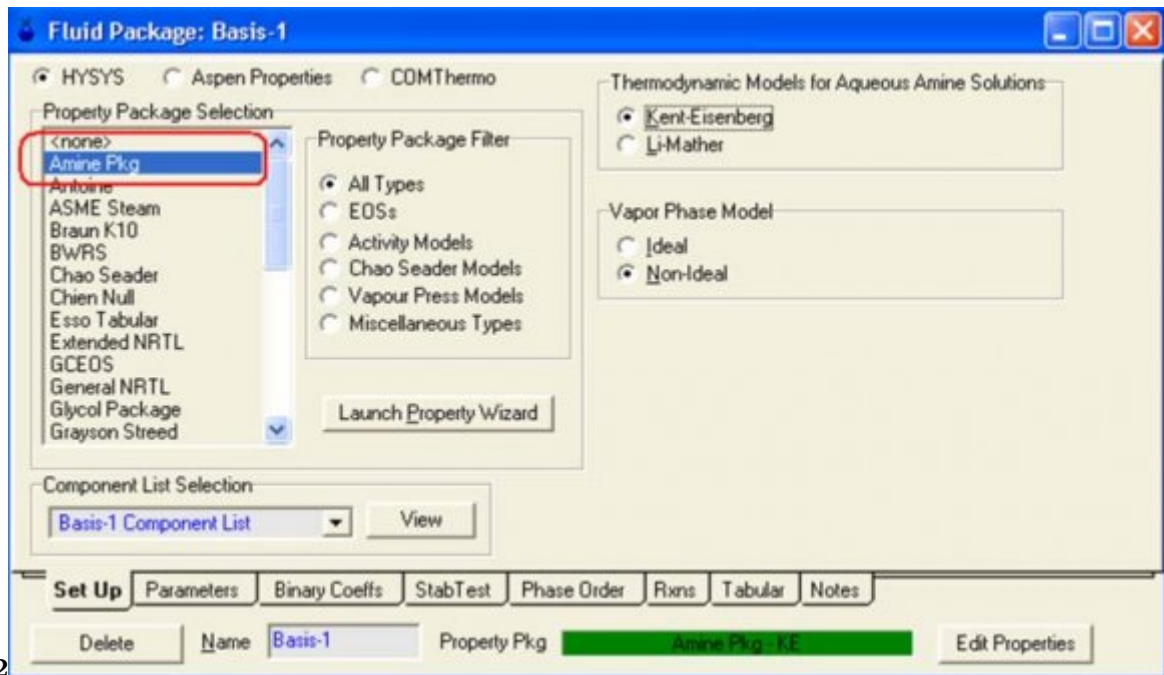


Figure 2:

Component	Mole%	RMM	Kmole/hr	Kg/hr	Mole%
H ₂ S	5.38	34.076	288.03426	9815.056	5.372849
CO ₂	4.48	44.01	239.85009	10555.8	4.474045
N ₂	0.11	28.02	5.8891764	165.0147	0.109854
CH ₄	63.35	16.02	3391.6302	54333.92	63.26579
C ₂ H ₆	13.9	30.07	744.17775	22377.42	13.88152
C ₃ H ₈	6.03	44.09	322.83394	14233.75	6.021985
i-C ₄ H ₁₀	1.36	58.123	72.811636	4232.031	1.358192
n-C ₄ H ₁₀	2.44	58.123	130.63264	7592.761	2.436757
i-C ₅ H ₁₂	1.03	72.15	55.144106	3978.647	1.028631
n-C ₅ H ₁₄	0.73	72.15	39.082716	2819.818	0.72903
C ₆ H ₁₄	1.19	86.177	63.710181	5490.352	1.188418
H ₂ O	-	18	7.1258541	128.2654	0.132922
Total	100		5360.9226	135722.8	100

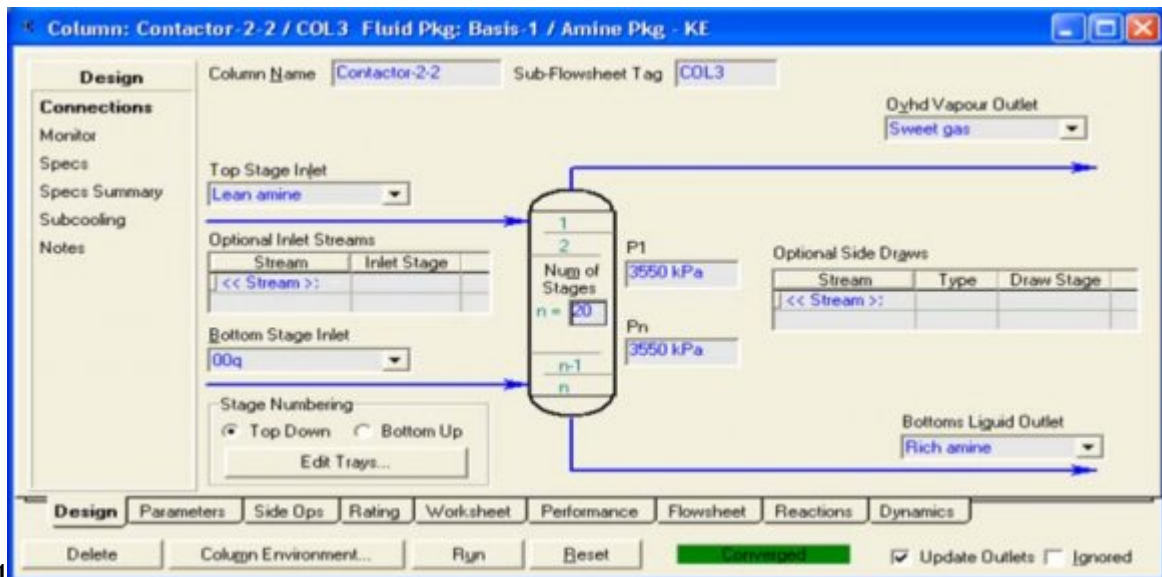
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Figure 3: Table 2 :



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Figure 4: Figure 2 :



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Figure 5: TheFigure 3 :Figure 4 :

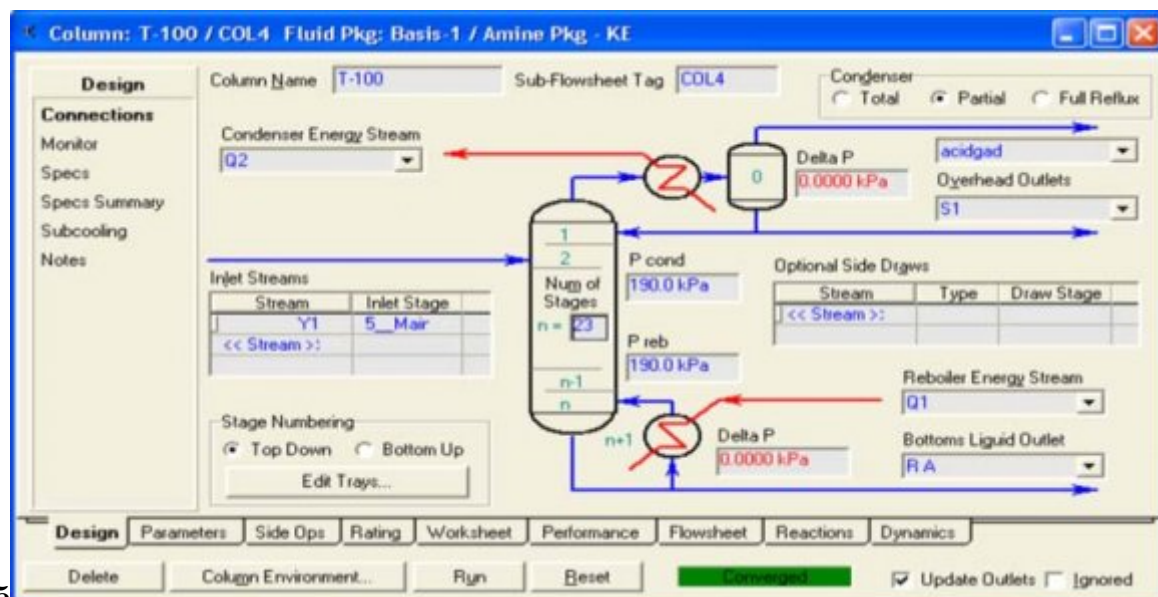
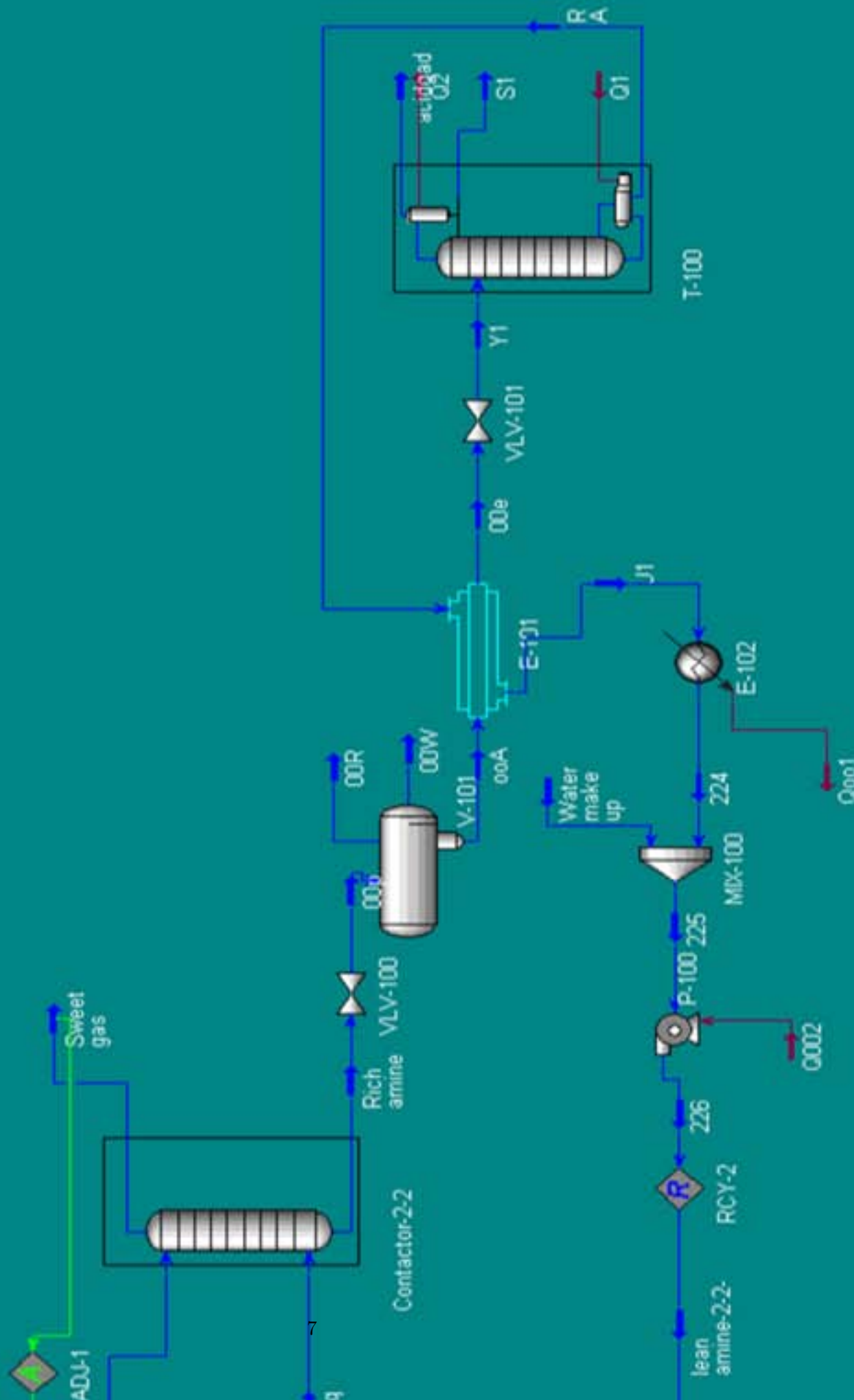
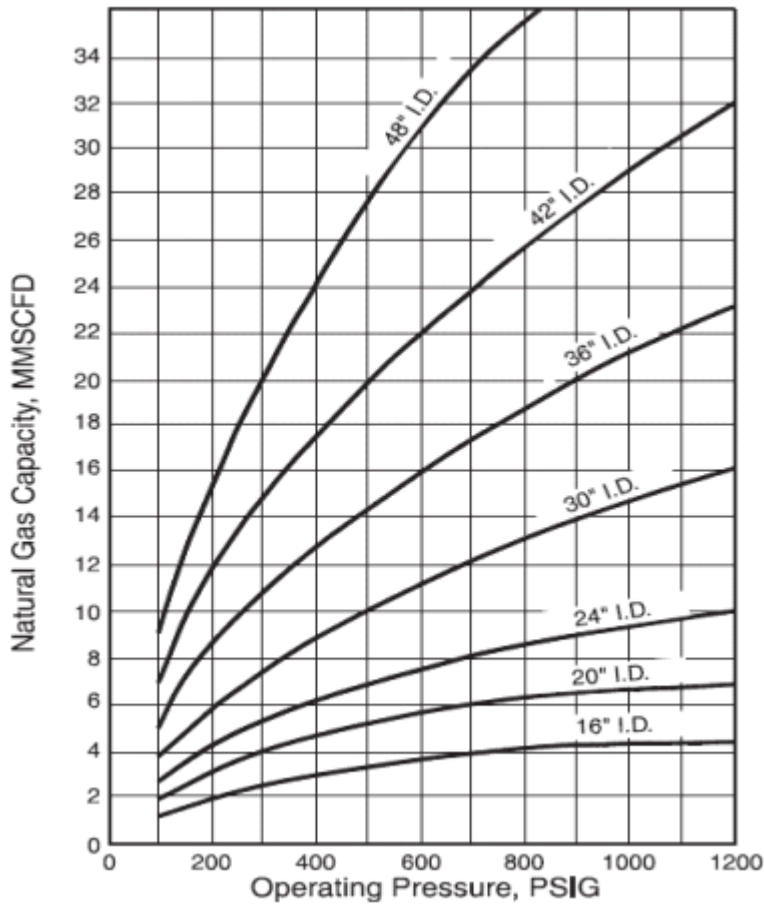


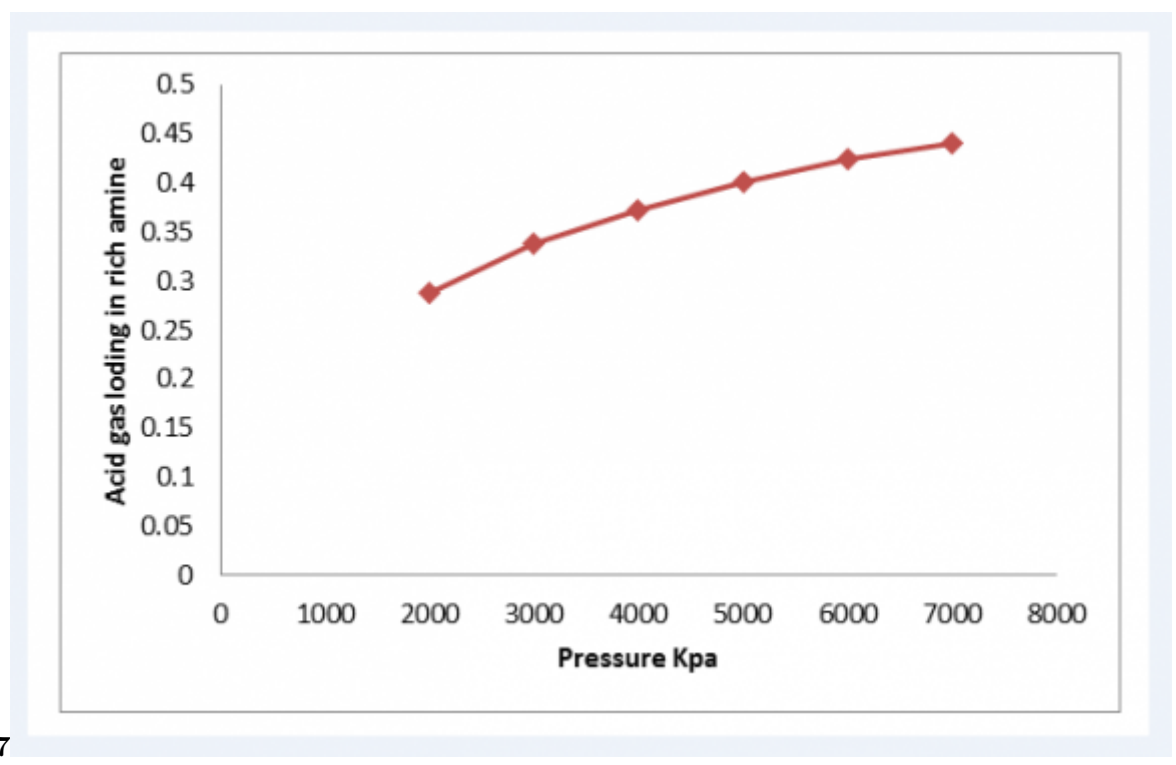
Figure 6: Figure 5 :





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Figure 8: Figure 6 :



7

Figure 9: Figure 7 :

1

		Component	Mole%
Case Study data		H ₂ S	5.38
		CO ₂	4.48
		N ₂	0.11
		CH ₄	63.35
		C ₂ H ₆	13.9
		C ₃ H ₈	6.03
Flow rate	120,000 stdm ³ /hr		
NG density	0.65 Kg/m ³	i-C ₄ H ₁₀	1.36
Gas SG	0.67	n-C ₄ H ₁₀	2.44
Pressure	7000 K.pa	i-C ₅ H ₁₂	1.03
Temperature	38 C°	n-C ₅ H ₁₄	0.73
Max. Ambient temperature	38 C°	C ₆ H ₁₄	1.19

Natural gas water content can estimate by using McKetta-Wehe Chart [3]. Therefore, water content is about 1000Kg/MMstd.m³ = 128.265 Kg/hr. Now, the new Natural gas composition could be calculated and summarized in table (2):

Figure 10: Table 1 :

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