

1 Impact of Flow Rate and Bed Height on the Fixed Bed  
2 Adsorption of Methylene Blue Dye on to Sphagnum  
3 Cymbifolium(Moss)

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8 **Abstract**

9 Objective: This work is aimed at expanding the field of application of natural biomass for the  
10 treatment of dye waste effluents. It is equally aimed at determining the dependency or  
11 otherwise of the effect of flow rate and bed height on the fixed bed adsorption of methylene  
12 blue dye onto Sphagnum cymbifolium (moss). Methods: The biomass was characterized by  
13 scanning electron microscopy (SEM) for the determination of the morphology of the biomass.  
14 The screened biomass samples were characterized at 1000X magnification, 500X magnification  
15 and 250X magnification respectively for their surface morphologies. This was done using a  
16 scanning electron microscope (FEI-inspect/OXFORD INSTRUMENTS-X-MAX), which was  
17 equipped with an energy dispersive x-ray (EDAX) spectrophotometer employed for the  
18 elemental composition analyses. It was equally characterized with Fourier transformed  
19 infrared spectroscopy (FTIR) before and after adsorption to ascertain the functional groups  
20 responsible for the adsorption. This was done using a Fourier Transformed Infrared (FTIR)  
21 spectrophotometer (Perkin Elmer, England) in the wavelength range of 350-4000nm.  
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29 before and after adsorption to ascertain the functional groups responsible for the adsorption.  
30 This was done using a Fourier Transformed Infrared (FTIR) spectrophotometer (Perkin  
31 Elmer, England) in the wavelength range of 350-4000nm.

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33 **Index terms**— bio-sorption, sphagnum cymbifolium, sem, fixed bed.

34 Abstract-Objective: This work is aimed at expanding the field of application of natural biomass for the  
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## 4 III. CHARACTERIZATION OF THE BIO-SORBENT

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43 to ascertain the functional groups responsible for the adsorption. This was done using a Fourier Transformed  
44 Infrared (FTIR) spectrophotometer (Perkin Elmer, England) in the wavelength range of 350-4000nm.

45 Results: Results for the biomass morphology through the scanning electron microscopy (SEM) showed the  
46 presence of some pores. These pores represent sites where dye molecules could be trapped in the course of  
47 adsorption. The results from the Fourier Transformed Infrared Spectroscopy (FTIR) before adsorption revealed  
48 the presence of five functional groups.

49 The functional groups include O-H or N-H, C-H, C?N or C?C, C=O, or C=C and benzene. However, after the  
50 adsorption, it was found that the functional groups that were responsible include C-H, C?H and C?C. Within the  
51 level of experimental consideration, it was found that the rate of adsorption was dependent on flow rate and bed  
52 height. An increase in flow rate and bed height led to a corresponding increase in the value of  $q_e$ . Conclusion:  
53 From the results obtained, it is seen that methylene blue dye can adsorb on to Sphagnum cymbifolium (moss)  
54 through the fixed bed process. Also, within the limit of experimental consideration, that the adsorption of  
55 methylene blue dye onto Sphagnum cymbifolium (moss) through the fixed bed technique is flow rate and bed  
56 height dependent. In each of the analyses, three different experiments were performed and the mean values  
57 reported with their standard deviations.

## 58 1 INTRODUCTION

59 any industries such as plastic, dyestuffs, textiles and inks, use dyes to color their products, and also consume  
60 substantial volumes of water. Due to their good solubility, synthetic dyes are common water pollutants. The  
61 presence of very small amounts of dyes from waste waters before it is discharged into the environment. Adsorption  
62 techniques are proved to be an effective and attractive process for the removal of nonbiodegradable pollutants  
63 (including dyes) from waste waters [1]. Most commercial systems use water because it has excellent adsorption  
64 ability. But, its widespread use is limited due to high running cost. Due to that, many low cost adsorbents, and  
65 waste materials from industry and agriculture have been proposed by several workers []. These materials do not  
66 require any expensive additional pre-treatment step and could be used as adsorbents for the removal of dyes from  
67 solution.

68 Some researchers reported the use of plant leaf biomass to adsorb heavy metals from solutions [2]. Resh  
69 water algae Pithophore sp; was studied by Kumar in finding out its bio-sorption properties on to malachite  
70 green (a cationic azo dye) [3]. This work is carried out with the view of expanding the field of application of  
71 natural biomass for the treatment of dye waste water through the fixed of application of natural biomass for the  
72 treatment for the treatment of dye waste water through the fixed bed technique. Also, it is aimed at determining  
73 the impact of flow rate and bed height on the fixed bed adsorption of methylene blue dye on to Sphagnum  
74 cymbifolium (moss). Since such as in-depth consideration has not been done on this biomass, the information  
75 obtained will add to the expansion of knowledge in this area.

## 76 2 II.

## 77 3 MATERIALS AND METHODS

78 The methylene blue dye, calcium chloride, distilled water and other necessary reagents used in this work were  
79 obtained from qualikem laboratory, owerri, Nigeria. The Sphagnum cymbifolium (moss) used was obtained from  
80 Ikorodu area in Lagos, Nigeria which is located within the following co-ordinates 6.6194°N and 3.5105°E. This  
81 sample was identified at the department of Crop science at the Federal university of technology, M lobal Journal  
82 of Researches in Engineering ( ) Volume Xx XI Issue II Version I J Owerri, Nigeria with the voucher specimen  
83 number of FUT/CR/005/17. The biomass was washed severally with distilled water to remove any dirt from it.  
84 The washed biomass was air dried for ten days until a constant weight was obtained. The biomass was grinded  
85 with a new sonic domestic blender to avoid any form of contamination. It was further screened using 600-800  
86 micron sized sieves and stored in air tight containers ready for adsorption.

87 This methods and techniques employed in these analyses are the standard methods which have been used by  
88 other researchers [4].

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90 The surface structure and morphology of the Sphagnum cymbifolium (moss) was characterized at 1000X  
91 magnification, 500X magnification, and 250X magnification, respectively. This was done using scanning electron  
92 microscopy (SEM) (FEI-Inspect oxford instrument-x-max), which was equipped with an energy dispersive x-ray  
93 (EDAX) spectrophotometer employed for elemental composition analysis.

94 The biomass sample was further characterized for their fundamental functional groups before and after  
95 adsorption experiment using Fourier Transformed Infrared (FTIR) spectrophotometer (Perkin-Elmer, England)  
96 in the wavelength range of 350-4000nm using KBr powder and fluka library for data interpretation.

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## 5 a) The Fixed Bed Set Up

The fixed bed was set up by packing wire gauze, glass wool, glass beads, glass wool, biomass and glass wool in that order in a graduated condenser. Then a dye solution of a known concentration and pH pressurized from down to top where a known amount of the bio-sorbent is placed with a peristaltic pump (CHEM-TECH model X030-XB-AAAA365, China). Subsequently, a sample was collected for u.v analysis in a u.v spectrophotometer (CAMPEC M 106 Model, England) by monitoring the absorbance already determined for methylene blue dye at 600nm. The variables investigated include the effect of flow rate and bed height.

## 6 b) Effect of Flow Rate on Adsorption

Experiments were performed at different flow rates of 2m<sup>3</sup>/s, 30m<sup>3</sup>/s and 40m<sup>3</sup>/s respectively, while keeping constant a bed height of 1x10<sup>-2</sup> m, 40mg biomass dose, 90mg/L and a pH of 4. The dye solution was subjected to pass through the column already prepared using the peristaltic pump. The samples collected were subjected to u.v analysis for absorbance values were converted to concentration by the use of Beer Lambert law. Similar experiments were carried out in triplicates and the mean values and standard deviation reported.

## 7 c) Effect of Bed Height on Adsorption

Experiments were conducted at different bed height of 4x10<sup>-2</sup> m, 5x10<sup>-2</sup> m and 6x10<sup>-2</sup> m while keeping constant a flow rate 10m<sup>3</sup>/s. 90mg/L dye solution, pH of 4 which is the pH of maximum adsorption for methylene blue dye. The dye solution was subjected to pass through the column already prepared using the peristaltic pump. The samples collected were subjected to u.v analysis for absorbance measurements at 600nm. Subsequently, the absorbance values were converted to concentration by the use of Beer Lambert's law. Similar experiments were carried out in triplicates and mean values and standard deviations reported. The SEM micrographs of Sphagnum cymbifolium (moss) showed the presence of unevenly dispersed cavities on the surface of the biomass. These cavities provide sites where the molecules of the dye could be trapped in the course of the adsorption. The SEM micrographs of (X250), (X500) and (X1000) magnifications are shown in fig. ??, 3 and 4 respectively.

## 8 NOTE: The amount of dye adsorbed per gram biomass

Similar cavities have been discovered by other researchers [5]. The FTIR Spectrum of Sphagnum cymbifolium (moss) after adsorption as shown in fig. 6 above was used to ascertain the functional groups that were responsible for the adsorption. After the adsorption, there were some bond displacement of the original peaks indicating the functional groups that were responsible for the adsorption reaction. The displacements occurred at 2925.71nm and 2363.51nm which corresponds to C-H, C?N and C?C functional groups. Although, the intensity of the peaks greatly decreased after the adsorption, the functional groups on the biomass did not disappear totally during the biomass characterization after the adsorption. As could be seen from fig. 7, an increase in flow rate caused a corresponding increase in the q<sub>e</sub> values within the range of experimental consideration. A similar effect was reported by other researchers [6]. This could be due to increase in the force of attraction between the dye solution and the biomass surface area. Figure 8 shows the effect of bed height on the quantity of the methylene blue dye adsorbed onto the biomass (q<sub>e</sub>). The q<sub>e</sub> values for the biomass increased with a corresponding increase in the bed height within the range of experimental consideration. The result indicates that, the longer the bed height, the higher the q<sub>e</sub> values. A similar situation has been reported in similar investigations [7]. This could be attributed to the longer time of interactions between the biomass and the dye solution.

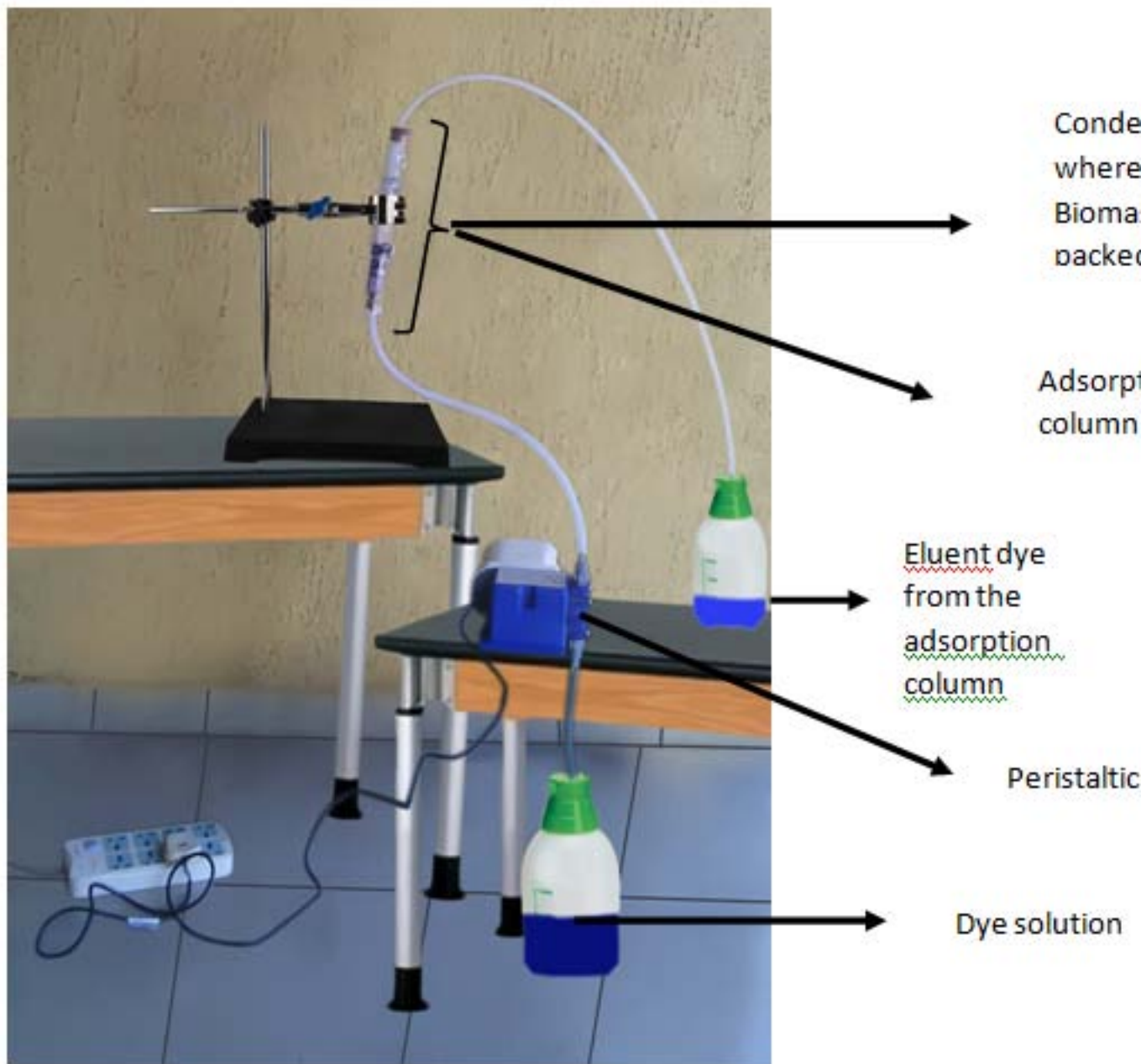
V.

## 9 CONCLUSION

The findings of this research vividly reveal that methylene blue dye can be adsorbed on to Sphagnum cymbifolium (moss) biomass through the fixed bed process. Additionally, the two variables, flow rate and bed height can impact the adsorption properties of methylene blue dye on to Sphagnum cymbifolium (moss).<sup>1</sup>

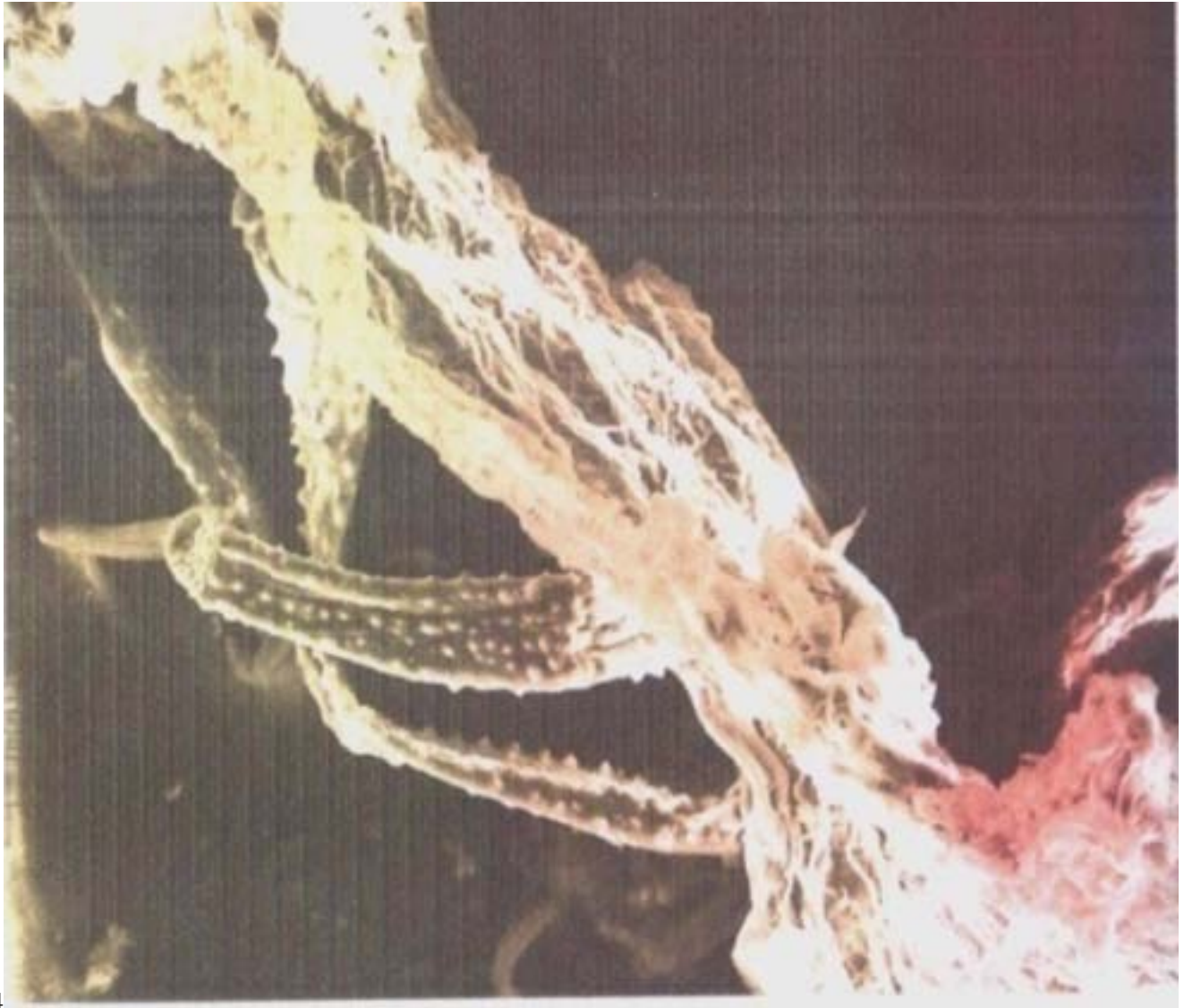
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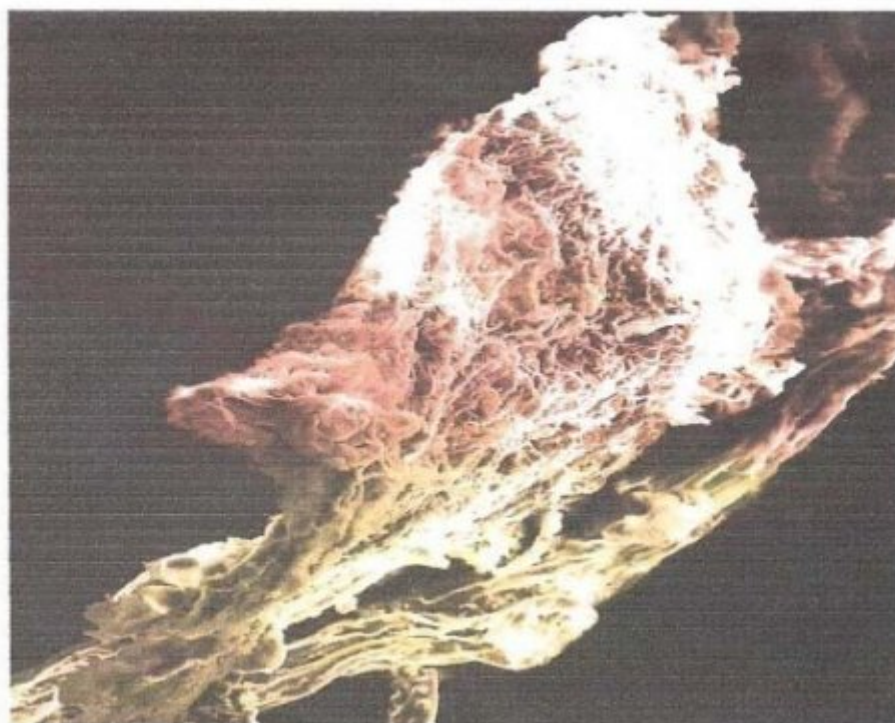
Figure 1: Fig. 1 :



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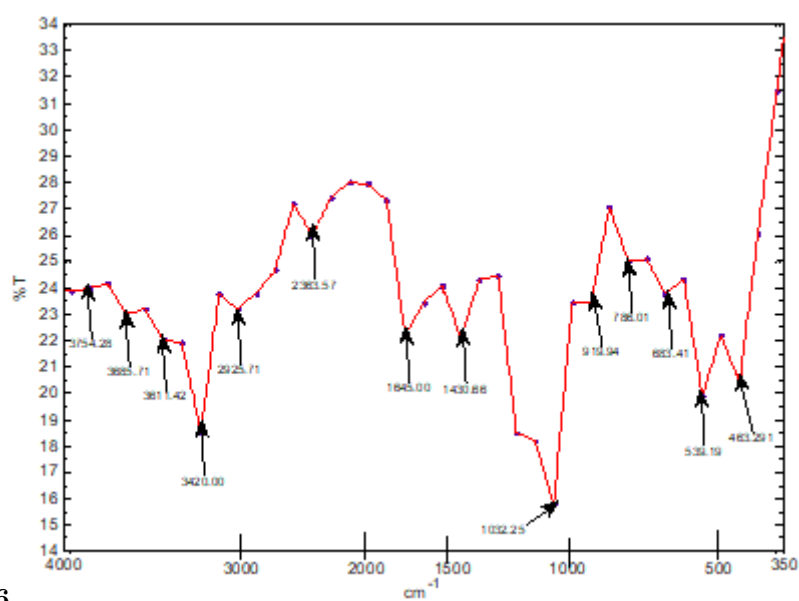
Figure 2: Fig. 2 :Fig. 3 :Fig. 4 :





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Figure 3: Fig. 5 :



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Figure 4: Fig. 6 :

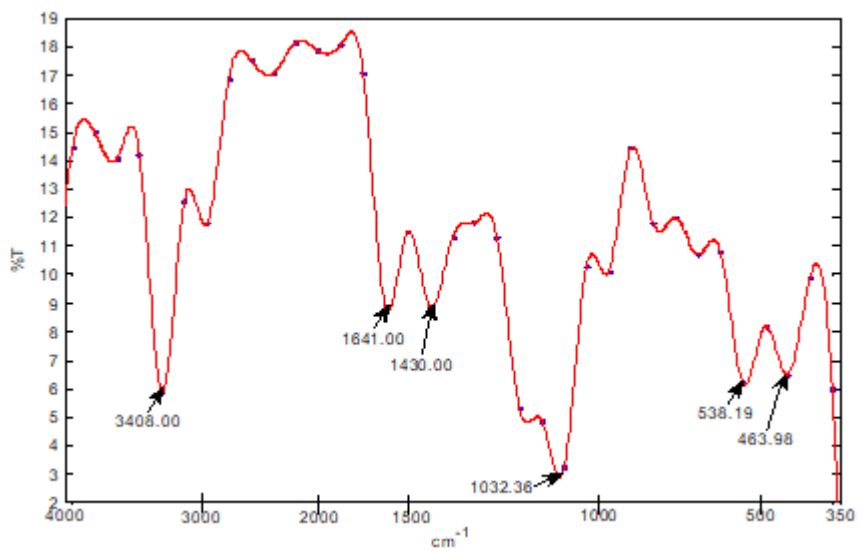
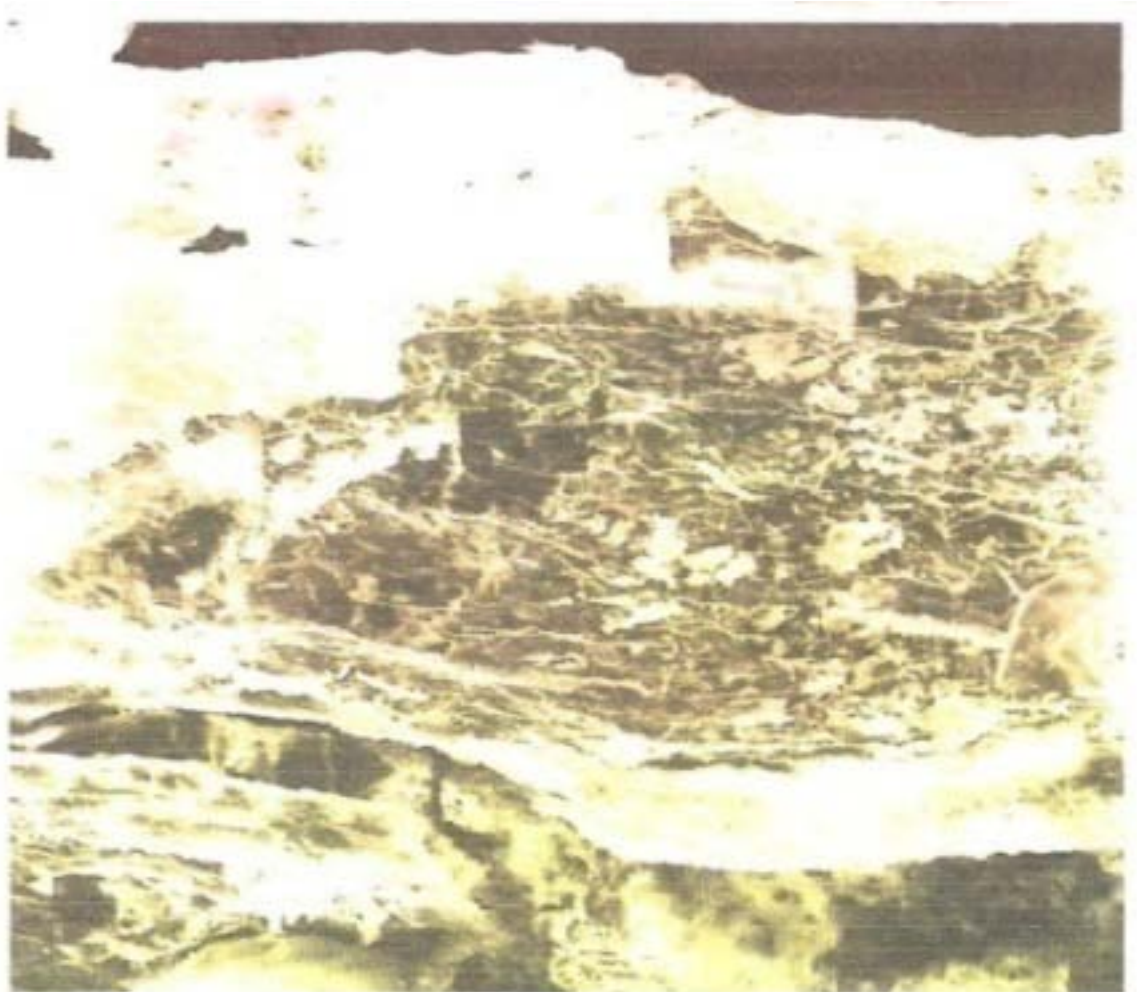


Figure 5: lobal



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Figure 6: Fig. 7 :

Figure 7:



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- 140 [Abraham ()] , R Abraham . *Environmental chemistry of chemical technology* 1993. Wiley. 8 p. .
- 141 [Gleason ()] ) *manual f vascular plants of North Eastern United states and adjacent Canada 2 nd ed*, Cronquist  
142 Gleason . 1991. p. .
- 143 [Cruz et al.] ‘2001) using in bed temperatures for visualizing the concentration front movement’. P Cruz , F  
144 Mendles , D Magaihaes . *Journal of chem. Engrn. Edu* 35 (27) p. .
- 145 [Namasiva and Sumithra] ‘2005) removal of direct 123 and methylene blue dye from water by adsorption onto Fe  
146 (III) / Cr(III) hydroxide, an industrial solid waste *Journal of Environ*’. C Namasiva , S Sumithra . *Management*  
147 74 p. .
- 148 [Khatttri and Singh ()] ‘Adsorption of basic dyes from aqueous solution by natural adsorbent’. S Khattri , M Singh  
149 . *Indian Journal of Chem. Technol* 1999. 6 (2) p. .
- 150 [Kumar et al. ()] ‘Adsorption of malachite green onto *Pitophora* SP. A fresh water algae / Equilibrium and  
151 Kinetic modeling’. K Kumar , S Savansan , V Ramanurhi . *Journal of process Biochem* 2005. 40 p. .
- 152 [Ghosh and Bhattacharyya ()] ‘Adsorption of methylene blue on Kaolinite’. D Ghosh , Bhattacharyya . *Journal*  
153 *of Appl. Clay Sci* 2002. 20 p. .
- 154 [Akzu ()] ‘Application of biosorption for the removal of organic pollutants: a review’. Z Akzu . *Journal of Process*  
155 *Biochem* 2005. 40 p. .
- 156 [Idika et al. ()] *Batch and fixed bed comparative study on the bio-sorption properties of Sphagnum cymbifolium*  
157 *(moss) on methylene blue dye, Bismarck brown y dye and indigo dye. The international journal of science*  
158 *and technoledge* 7, Digbo Idika , I Ndukwe Nelly , C , Ogukwe Cynthia , E . 2019. p. .
- 159 [Idika et al. ()] *Batch and fixed bed comparative study on the dye bio-sorption properties of Cedrus libani (Elizabeth*  
160 *leaf) on methylene blue, bismarck brown y and indigo dye. The international journal of science and technoledge*  
161 7, Digbo Idika , I Ndukwe Nelly , C , Ogukwe Cynthia , E . 2019. p. .
- 162 [Idika et al. ()] *Bio-sorption properties of Sphagnum cymbifolium (moss) on methylene blue, Bismarck brown*  
163 *y, and indigo dyes by the batch process. acta SATECH* 11, D I Idika , C A Ogukwe , E E Oguzie , A O  
164 Alishinloye , A Adewunmi . 2019. p. .
- 165 [El-Geundi ()] ‘Colour removal from textile effluent by adsorption techniques’. S El-Geundi . *Journal of water*  
166 *Res* 1991. 25 p. .
- 167 [Idika et al. ()] ‘Effect of flow rate and bed height on the fixed bed adsorption of methylene blue dye, Bismarck  
168 brown y and indigo dyes onto *Sphagnum cymbifolium* (moss)’. D I Idika , N A Ndukwe , C E Ogukwe . *IOSR*  
169 *journal of applied chemistry* 2020. 13 p. .
- 170 [Vadivehan and Vasanth ()] ‘Equilibrium Kinetics, Mechanism, and process design for the sorption of methylene  
171 blue onto rice husk’. V Vadivehan , K Vasanth . *Journal of Colloid Inter Sci* 2005. 286 p. .
- 172 [Gupa et al. ()] ‘Equilibrium uptake and sorption dynamics for the removal of basic dye (basic red) using low  
173 cost adsorbents’. V Gupa , . I Ali , M Suhas . *Journal of colloid interface Sci. P* 2003. p. .
- 174 [Guibal ()] ‘Interaction of metal ions with chitosan based sorbents -a review’. E Guibal . *Journal of separation*  
175 *and purification Technol* 2004. 38 p. .
- 176 [Crini ()] ‘Non conventional low cost adsorbents for dye removal’. G Crini . *Journal of Bioresour. Technol* 2006.  
177 79 p. .
- 178 [Robinson et al. ()] ‘Remediation of dyes in textile effluent: a critical review on current treatment technologies  
179 with a proposed alternative’. T Robinson , G McMullan , R Marchant , P Nigam . *Journal of Bioresour.*  
180 *Technol* 2001. 77 p. .
- 181 [Robinson et al. ()] ‘removal of dyes from a synthetic textile dye effluent by bio-sorption on apple pomace and  
182 wheat straw’. T Robinson , B Chandran , P Nigam . *Journal of water Res* 2002. 36 p. .
- 183 [Idika et al. ()] ‘Studies on the impact of flow rate and bed height on the fixed bed adsorption of methylene  
184 blue dye, bismarck brown y dyeand indigo dye onto *Cedrus libani* (Elizabeth leaf) biomass’. Digbo Idika ,  
185 Nelly Ndukwe , Cynthia Ogukwe , Adewunmi Aderike . *International journal of chemistry research* 2020.  
186 Aleshinloye Abimbola. 4 p. .
- 187 [Langmuir ()] ‘The constitution and fundamental properties of solids and liquids’. I Langmuir . *Journal of Am.*  
188 *Chem.Soc* 1916. 38 p. .
- 189 [Ai-Subu (200)] ‘The interaction of Cypress (*cupressus sempervirens*), cino chona (*Euclyptus longifolia*), and  
190 pine (*Pinus helepenses*) leaves on their efficacies for lead removal from aqueous solution’. M Ai-Subu . *Journal*  
191 *of Environ resource* 200. 6 p. .
- 192 [Amadurai et al. ()] ‘use of Cellulose based wastes for adsorption of dyes from aqueous solutions’. G Amadurai ,  
193 R Janig , D Lee . *Journal of hazard mater* 2002. 392 p. .