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1	Effects of Biomass Properties on the Performance of a
2	Gasifier/Genset System
3	Francisco Everton Tavares de Luna
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5	

6 Abstract

 $_{7}$ $\,$ Biomass can be considered one of the most important sources of energy in the world, because

⁸ it is: renewable; neutral in terms of green-house gases emissions; capable of replacing

⁹ conventional fossil fuels, among other factors. On the other hand, gasification is an efficient

¹⁰ process of turning available the chemical energy of biomass, with a relatively simple

¹¹ technology. In the present work a co-current open top downdraft gasifier is used, with an 8.5

12 kW thermal power capacity to fuel an 18 Hp Otto cycle engine coupled to an electric

¹³ generator. With this apparatus, it was possible to analyze the influence of some properties of

the fuel wood particles (size, density, moisture content and so on) on the efficiency of the

¹⁵ energy conversion process.

16

17 Index terms— biomass gasifier; gasifier/genset system; electricity generation.

Effects of Biomass Properties on the Performance of a Gasifier/Genset System Introduction t least five facts 18 underlie the understanding that biomass is the most important source of energy in the world, [1], [2], [3], [4]. [5] 19 [6], and they are based on the following: 1. It is a renewable fuel, [7], [8]; 2. It is neutral as regards the emission 20 of greenhouse gases, [1], [9], [10]; 3. It is capable of replacing conventional fossil fuels, [1] to [6]; 4. It is abundant, 21 [2], [3] and [6]; Its resources are found almost everywhere [11][12]. There are several biomass conversions, with 22 different characteristics and results [13]. The most efficient way to make the internal chemical energy of biomass 23 available is through the production of gas either by biochemical (fermentation) or thermo chemical (pyrolysis) 24 processes, the latter requiring more external energy, but with faster practical results [14]. 25

²⁶ 1 a) Biomass Gasifier and Gasification Process

As well known, depending on their characteristics (method of heating, gasification agent, pressurization, transport 27 processes, etc.) gasifiers may be classified into different types, [13], [15]. When the distinction is based on the 28 way biomass and the gas flow move, biomass gasifiers are conceived of as fixed bed (updraft, or downdraft), 29 fluidized bed, entrained flow, etc. The fixed bed gasifier with a fuel hopper top (also known as moving bed) is 30 the most common [16]. It has been preferred to the closed top gasifier, such as the Imbert gasifier (throated or 31 closed top gasifier). The reasons are: the fuel is easily fed; quick access to the instrumentation for needed control 32 measurements; air and biomass pass uniformly downward through the four zones (drying, pyrolysis, combustion 33 and reduction), avoiding excessive deviation from the local high average temperature; less trouble with channeling 34 or bridging events; the top zone may be easily and conveniently adjusted [15]. 35

Gasification agents may be air, steam, oxygen or CO 2. The fixed bed gasifier, also considered very suitable 36 37 for internal combustion engines, by reason of producing low tar content, [16], [17], is appropriate for small to 38 medium scale thermal applications [18]. Depending on the gasification agent flow direction, a gasifier may be 39 designated as countercurrent, cocurrent, cross flow, etc. Generally speaking, the cocurrent gasifier is used in small scale power generation and the air coming from nozzles set around the reactor zone, as well as from the 40 top (about 60 %) moves downward in the same direction as the produced gas (the poor gas). It is observed that 41 in co-current gasifiers air input rates regulate the fuel consumption rates [19]. On the other hand, the reactor is 42 simple to construct and generates a poor gas with low tar in its composition [20], [21]. 43

Particle size is one of the most recurrent independent variables appearing in almost all pyrolysis or devolatilization models through a non-dimensional number [22]. However, most pyrolysis studies do not make

reference to any non-dimensional number, see e.g. [23], [24], [25], [26]. Thus, considerations of the influence of the 46 fuel dimension on the gasifier functionality, mostly come from phenomenological results, allowing to enunciate 47 some statements such as: 1. Fine grained, or fluffy particles may produce gas flow difficulties inside the gasifier 48 body reactor [27], with considerable pressure drops over the reduction zone; 2. Disproportional large sizes can 49 give rise to bridging and channeling problems [4]; 3. Biomass particle size, as well as, its moisture content are 50 important factors affecting the combustion and heat recovery, especially if combustion is incomplete [22], [24] 51 and [28]; 4. The flame propagation speed, i.e., the rate of progress of the apparent flame zone, is dependent on 52 the particle size, as well as on the air supply rate, and the calorific value of the solid fuel, Shin et al. [29]; 5. A 53 reduction in the fuel particle size leads to a significant improvement in the gasification parameters, Hernandez et 54 al. [30]. 55

Not only should size, but also particle density be considered when the goal is to improve gasification results. In 56 fact, it is easy to notice that density often figures in the chemical kinetics and transport phenomena correlations, 57 where those fundamentals, as mentioned above, are necessary to help to describe the pyrolysis models [10], 58 [30], [31] and [32]. Huff [33] demonstrated the importance of size, shape, density, moisture, and wall furnace 59 temperature in the burning time of single pieces in fireboxes. 60

61 In reading the technical literature, we understand that the influence of the biomass particle size on the 62 gasification process has been extensively, theoretically or experimentally, studied. However, it should be noted 63 that most of the studies, experimental, or theoretical (models), take into account just isolated particles, [21], [22], 64 [28], [29], [30], [31], [32], [33].

It was only around 1920 that poor (producer) gas was used to fuel engines, Shrinivasa et al. [34]. In fact, the 65 petroleum shortage during World War II led to widespread applications of gas generation in the transportation 66 industries of Western Europe, La Fontaine et al. [35]. As mentioned by FAO [27], spark ignition engines can 67 be run on poor gas (producer gas) alone, and Diesel engines can be converted into full poor gas after being 68 submitted to some modifications, or run in a dual mode. The use of poor gas on internal engines, tar and 69 particulate contents have since been proved too great a hurdle. This fact motivated the IndianInstitute of Science 70 in Bangalore, see ??asappa et al. [36], to develop biomass gasifiers capable of cleaning and cooling the poor gas, 71 to be used in dual fuel mode (diesel/poor gas). In fact, the majority of poor gas application in engines uses the 72 dual mode, e.g. Shrinivasa et al. [34], Dasappa et al. [36], Sridhar et al. [37], Dasappa et al. [38], Kalina [39] 73 and Ghosh et al. [40]. Less frequent is the utilization of IC engines fueled just on poor gas: Raman et al. [41], 74 75 for example, used an engine designed to run on natural gas to operate on 100 % producer gas, and Gitano [42] 76 modified a gasoline two-stroke genset for operating on syngas (producer gas) from a biomass gasifier. The present work discusses the global efficiency of a system formed by a co-current, downdraft fixed bed 77

biomass gasifier, coupled to a genset, and an Otto Cycle engine to generate electricity. The biomass gasifier fuels 78 the genset with a hundred percent poor gas. The influence of some biomass properties, such as size, density and 79 moisture content on this overall process is analyzed. 80

2 II. 81

Producing the Poor Gas a) Dynamics of the gasifier reactor At least four stages are necessary for biomass 82 gasification: drying, pyrolysis, combustion and reduction. Being dependent on heat transfer properties, the 83 drying process, aside from the moisture and the ash content, may also depend, as already reported, on some fuel 84 (biomass) physical parameters, such as size, heat diffusivity, heat capacity, heat transfer coefficient, and thermal 85 conductivity. At the beginning of the process, there is evaporation inside the fuel, production of condensable 86 fractions with loss of water, which happens at temperatures above 100 o C. On the other hand, volatiles are 87 88 released at temperatures close to 140 o C. At the same time, steam escapes from the particles, causing fuel and 89 pores shrinkage, as well as the ending of the drying process. As the temperature increases, it is easy to detect the presence of CO 2 and CO, chiefly when cellulose is heated at 170 o C, Hill [43]. Generally speaking, pyrolysis or 90 release of volatiles have been considered as the first stage in gas production from biomass, Di Blasi [6]. The use 91 of thermo gravimetric analysis shows that all volatiles are released up to 500 o C, the lignin at this temperature 92 being completely thermally degraded. Tar, the product of destructive distillation, and ash in the reactor occur 93 at temperatures higher than 800 o C, Yoshikawa [44]. It is observed that the pyrolysis product will react at high 94 temperatures, 700 to 1500 o C for existent gases, chiefly for external O 2, in the combustion zone, where secondary 95 reactions generally occur. During this process conversion of residual char is detected, presenting much slower 96 reaction than the oxidation process, Basu [45], determining the overall gasification efficiency. Finally, as particles 97 move into the reduction zone, they become smaller due to the consumption of the char by surface reactions. 98 99 It is also in this zone that the char particles act as reducing agents for the remaining gaseous compounds, De 100 Santanu [46], forming the poor gas, basically a mixture of H 2, CO and CO 2. Year 2020 to produce electricity. 101 The gasifier reactor 0.90 m long with internal and external diameters of 0.16 m, and 0.18 m, respectively, has 102 the annular space filled with vermiculite. The genset parts are: an original gasoline VANGUARD V-Twin, 2 cylinders, 18-hp Otto cycle, adapted to run on poor gas. and a generator from Toyama (model TG2500MX), 103 single phase, 220 V and 60 Hz. 104

A resistive charge simulator with eight electric resistances is capable of testing electric powers up to 2.4 kW. An 105 electric energy analyzer from HIOKI is used to evaluate the frequencies, current, and the electric power produced 106 by the genset. 107

Gases emissions (CO, HC, NO x and CO 2) and the lambda factor are evaluated by means of an Alphatest vehicular gas analyzer.

110 A thermocouple, K type, is used to evaluate the exhausted gases temperature.

111 3 c) The Biomass

Four different types of waste wood material, brought from the university campus dump and cut into uneven cubic pieces, originated the four different biomass samples, characterized by their four different edges (The first, third and fourth samples were from the species Tabebuia heptaphylla, and the second from Ceasalpinia echinata). On average, the edge and the cubic volume of the samples (1 to 4) were respectively, 13 mm (2; 197 mm 3), 16 mm (4,096 mm 3), 20 mm (8,000 mm 3) and 27 mm (19,683 mm 3). For each one of the tests, the gasifier ran with just one kind of sample.

118 The moisture content of each one of the four samples was determined experimentally in triplicate.

For the analysis of the biomass sample results, a proximate analysis, using the ASTM E-1131 Standard Test Method for Compositional Analysis by Thermo gravimetry was also conducted in triplicate. For these tests, 30 mg of each sample with an average diameter of 100 mm, was brought to a 100 mL.min -1 gas flow (N 2 and

122 synthetic air), using different temperature levels.

¹²³ 4 d) The low heating value of the poor gas

As mentioned by Reed et al. [17], the gas heating value of raw producer gas containing significant condensable 124 volatiles (tars) is difficult to measure, since the measurements are made at room temperature after the tar has 125 been condensed. Generally speaking, in the technical literature, we find different average values. For Reed et 126 al. [17], the lower heating value, LHV, of the producer gas, situates between 5-7 MJ.Nm -3; Barrio et al. [47] 127 4.85 MJ.Nm -3; Albertazzi et al. [48], 5 MJ.Nm -3; Kaupp et al. [49] between 4 and 6 MJ.Nm -3. There 128 are, however, two publications, Yoshikawa [44] and Garcia [50], that show the plot of the LHV of the poor gas 129 given in function of the percentage of carbon monoxide by volume of poor gas. Based on this set of scattered 130 points, Rumão [51], using a curve fitting process, determined Eq. (??), which produced a Pearson's correlation 131 coefficient equal to 0.9379, with a standard deviation of ? p = 0.975 MJ.Nm -3. The correlation, see Eq. (??), 132 gives the LHV of the poor gas in terms of the percentage of CO by volume of poor gas, as MJ.Nm -3. (Typically, 133 in the poor gas composition, for hydrogen and carbon monoxide, it is 19 ± 1 % H 2 and 19 ± 1 % CO. Therefore, 134 in Eq. (??) the effect of H 2 was replaced by the one of CO by just altering its coefficients); LHV poor gas = 135 -0.004738.(%CO) 2 + 0.3149.(%CO) -0.1057 MJ.Nm -3 (1)136

e) Efficiency of the system gasifier/genset Equation (2) was used to evaluate the efficiency of the system (gasifier/genset)? sys = $p \in M$?b .LH V bio 100 %(2)

Where p e is the generated electric power, W; M ?b is the evaluated mass flow used to feed the gasifier, kg/s;
LHV bio is the average biomass low calorific value, J/kg, which was determined experimentally in triplicate.

¹⁴¹ 5 f) Determining the efficiency of the internal combustion ¹⁴² engine coupled to the genset

Since the final efficiency of the system depends on the efficiency of its elements, a series of experiments was made to determine the efficiency of the internal combustion engine coupled to the genset. The engine efficiency was evaluated using its original fuel, i.e. gasoline, choosing the better valve clearance to guarantee the maximum efficiency. After correcting the pressure rate of the engine running with poor gas, a new evaluation of the engine efficiency was determined, using Eq. (??)? e = P gen P gas 100 % (3)

where, P gen is the power generated, W. P gas , the power liberated by gasoline, whence, P gas = m?LHV gas(4)

m? being the gasoline volumetric flow rate, m 3 /s, and LHV, the lower heating value, J/kg (admitted as being 42680 kJ/kg).

¹⁵² 6 g) Running the system

Figure ??: The Y shaped mixture air/gas controller First the biomass inside the reactor is ignited with a gas 153 torch burner. Within ten minutes, the gasifier flare is lit. The flare intensity and color start changing as well 154 as the CO level of the poor gas. To start running the engine, the CO level must go up to 10 %. To guarantee 155 an approximate stoichiometric mixture of air/poor gas there is an Y shape mixing apparatus, see Figure ??. A 156 157 load bank resistor (power range from 0.7 kW to 2.2 kW), was used to simulate the resistive load of the generator. 158 Having stabilized the engine, (indicated by a close value of the 60 Hz frequency, as registered by the control equipment), the electrical resistances start being loaded, and all the data (power, biomass consumption, gas 159 composition, elapsed running time, etc.) are registered. The biomass consumption is checked by means of a 160 digital scale, considering that at the beginning of the tests, the biomass fills the fuel hopper to its maximum 161 level. During the operation, new quantities of weighted biomass (in kg) are used to feed the gasifier, and the 162 elapsed time is registered. The composition of the poor gas as well as that of the exhausted gases is evaluated 163

using a Discovery G4 vehicle gas analyzer, fromAlfatest. The whole procedure is repeated for each of the four samples of wood pieces.

166 **7** III.

¹⁶⁷ 8 Results and Discussion

¹⁶⁸ 9 a) The Biomass Moisture Content and Density

Table 1 shows the moisture content determined experimentally for the four biomass samples used to feed the 169 gasifier. Table ?? presents the average density, experimentally determined, of the four wood samples. The values 170 of the moisture content in Table 1 are all very similar, having magnitudes lower than 10.2 %. (To avoid producing 171 lower biomass heating values, the moisture content should not be higher than 15 %, [52]). 3 presents the results 172 of the proximate analysis of the four different biomasses, using the ASTM E-1131 Standard Test Method for 173 Compositional Analysis by Thermogravimetry. It shows that all the samples present high percentage of volatile 174 matter, facilitating the conversion and the upgrading of the fuel, Digman et al. [54]; As a result of its smallest 175 percentage of volatile matter, sample 4 presents the highest percentage of fixed carbon (FC). Thus, consonant 176 with its FC magnitude, its HHV is larger than those of the other samples, which show similarly smaller values. 177 It should be remembered that fixed carbon is the solid carbon of the biomass which remains in the char after it 178 has been submitted to the devolatilization and pyrolysis processes, as pointed out by Basu [45]. On the other 179 hand, the smallest percentage of ash was found in sample 3. In terms of moisture we canconsider that all samples 180 have similar contents. 4 shows the temperature registered inside the reactor, in the drying, pyrolysis, combustion 181 and reduction zones. As expected, the temperatures mount till the combustion zone, declining at the reduction 182 zone, and depending on the biomass, the temperature changes for each of the zones in question. This behavior 183 directly influences the percentage of CO, CO 2 and O 2 generation, see Figure 3. It shows the four types of 184 biomass CO, CO 2 and O 2 levels, at the engine's maximum power. 3, the CO level percentage increases as 185 the sample volume mounts. This trend repeats for the CO 2 percentage levels all along most part of the curve. 186 It seems that the size of the sample interrupts this tendency. On the other hand, the O 2, by reason of the 187 188 CO 2 and CO gases formation, is the only curve that goes down continuously, presenting an almost fixed slope. 189 e) The Poor Gas LHV as regards the electric power generation Figure 4 presents the CO, and O 2 percentage 190 as regards their biomas s densities. The tendency lines of gases CO, and O 2 present, as expected, an inverse behavior to CO 2 lines. Comparison between the curves in Figures 3 and 4, given the fact that the formation 191 of the gases CO and CO 2 is enhanced by the increase in temperature, indicates that the flame zone intensity 192 is much more limited by particle density, than by particle size. This fact is supported by the data in Tables 2 193 and 4, which show that lower densities correspond to higher temperatures in the pyrolysis zone. In consequence, 194 the O 2 behavior in Figure 4, is characterized by an increasing tendency, as opposed to what occurs in Figure 195 3. Figure 5 shows the heating value curves of the poor gas as a function of the electric power generation for 196 the four samples. Differently from what happens with the majority of gasifiers, which use a blower to improve 197 combustion, the enhancement of the flame inside the gasifier is mainly done by engine aspiration, acting as a 198 driving force for gasification. As mentioned by Shin [29] the biomass size, as well as its calorific value may also 199 influence the flame propagation speed. In Figure 5 we can see that considering the full range of variation of the 200 electric generated power, the lowest LHV average is related to the samples having the highest average densities 201 202 -1073.435 kg.m -3 and 862:444 kg.m -3 -i.e. samples 1 and 3, respectively (see Table ??). Whereas sample 4 (? = 743.358 kg.m -3), with the lowest average density and the largest LHV value, is the only one to show a 203 continuous rising of the LHV. On the other hand, the second largest LHV value is produced by sample 2 (? = 204 748.238 kg.m -3), which shows a rapid evolution of the generated electric power, but rapidly falls after reaching 205 1.7 kW. It should be noted that samples 4 and 2 present both the lowest density and volatile matter, see Table 206 3, while sample 4, shows the largest physical volume. f) Biomass Specific Consumption Figure ?? presents the 207 biomass specific consumption in terms of the electric generated power, for the four different sizes of biomass. We 208 see that, in general, the specific consumption of the biomass decreases with the increase of the generated power 209 level, the lowest consumption being achieved by sample 4 type (considering the whole range of electric power 210 generated), and sample 3 coming next (their densities are respectively 743.358 kg.m -3 and 862.444 kg.m -3). 211 For the electric power ranging from 0.9 kW to 2.2 kW, the consumption raised on average, 2.5 kg/kWh, when 212 213 the gasifier was fueled with sample 1 type (? = 1073.435 kg.m -3). When the system is running with sample 4 biomass type, (? = 743.358 kg.m - 3) the consumption is the smallest, as compared with the other biomass 214 types. 215

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Figure ??: Biomass specific consumption g) Efficiency of the system Gasifier/ Otto Cycle engine/Generator Figure 7 presents the plot for the system (gasifier/genset) efficiency, see Eq. (2), in terms of the generated electric power. It shows that from the smallest power up to 1.8 kW, no matter the sample, the efficiency of the system tends to increase. From this point on, in three of the cases, the curves show a slight decrease as the

- electric power increases. The highest efficiency (11.99 %) results from the use of sample 4 biomass (? = 743.358 kg.m -3), when the electric power reached 1.85 kW. In this connection, Tinaut et al. [55] using a onedimensional
- stationary model of biomass gasification to study the effect of the biomass particle size on the gasification process

in a downdraft fixed bed gasifier, showed that the maximum efficiency was achieved with a smaller particle size.

In their case, the model was validated experimentally in a small-scale gasifier by comparing the experimental temperature fields, biomass burning rates with predicted results. However, the biomass density was not taken

into consideration. In another model developed by Thunman et al. [24], concerning solid fuel conversion in a

229 grate furnace using a fixed bed fuel bed, they concluded that particle density has small influence on the conversion

rate, but noted that the particle size influenced the combustion behavior. In our case, however, small density

has shown to have a beneficial influence on the various aspects of the gasifier, i.e. on its behavior and on the \mathbf{E}

electricity production system, see Figure ??. ??), gave as result ? e = 16.87%, to generate 2 kW electric power. And as we have seen, the maximum efficiency of the system (gasifier/genset), ? sys , for generating electricity

was 11.9 %, which may be considered low. If the efficiency of the genset, ? gens , running on its maximum power

is of 13.5 %, i.e. 80 %, of the power determined when run on gasoline, it becomes evident, from Eq. (??), that

the gasifier efficiency, ? g , is, in fact, 88.1 %, ? g = ?sys ?gens (5) IV.

237 **12** Conclusions

The dissimilar curves in Figures 3 and 4, are an indication that we cannot analyze gasification performance referring just to biomass size, as Hernández et al. [30] did. Therefore, because of an existing correlation between biomass size and density, we can conclude, see Figure 3, that the larger the sample, the greater the CO percentage. Concerning the CO 2 formation, it seems that there is a sample size limit (associated with a determined density value), when its production decreases caused by flammable shortage.

The most remarkable fact registered in the several tests concerning sample 4 (? = 743.358 kg.m - 3) is that 243 it allows the maximum temperature of the reactor combustion zone. Analyzing its average figures of moisture 244 content, density, and higher heating value, and comparing them with those of other samples, it is clear that 245 sample 4 reunites the suitable property values to guarantee the adequate conditions for generating electricity, 246 with the smallest biomass consumption. In other words, it shows the best effective energy efficiency among all 247 the samples. It is also possible to conclude that the smaller the density, the slower the specific consumption, see 248 Figure ??. Consequently, lower density helps the gases residence time raise, enabling a more efficient gasification, 249 as indicated by the decreased concentration in O 2 , see Figure 4. According to Billaud et al. [56], CO 2250 formation occurs from combustion reactions and is directly bound up with the amount of O 2. As a consequence 251 of higher temperatures, there is an elevation in carbon monoxide concentration, a flammable gas, cf. Yin et al. 252 [57]. It should be mentioned that similar results were obtained by Feng et al. [25], in studying a catalytic steam 253 gasification of biomass. The only divergence is the behavior of CO 2, which decreased in a certain portion of the 254 255 curve, due to the increase of the volume sample, as well as of its density. On the other hand, it should be noted that, given the HHV function of the CO level, the higher heating value of the poor gas made sample 4 biomass 256 257 (? = 743.358 kg.m -3), the only one capable of offering the system maximum efficiency in generating electric power. 258

Considering both the maximum efficiency of the system, and the efficiency of the engine running with poor gas, we can conclude that the gasifier efficiency with maximum power is about 88.1 %, undoubtedly, a standout figure, Ptasinsky [58].

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Figure 1: Figure 1 :



Figure 2: 2020 Table 2 :

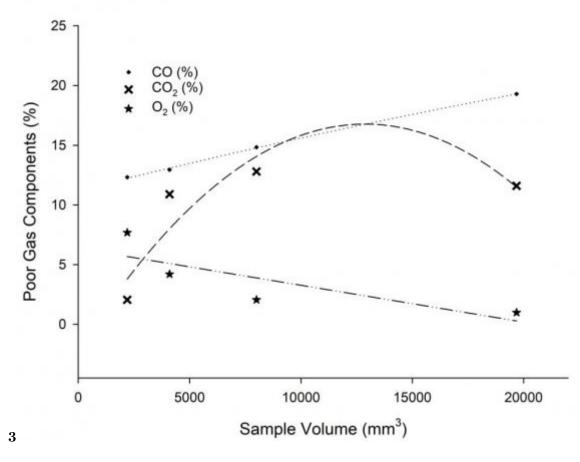


Figure 3: Figure 3 :

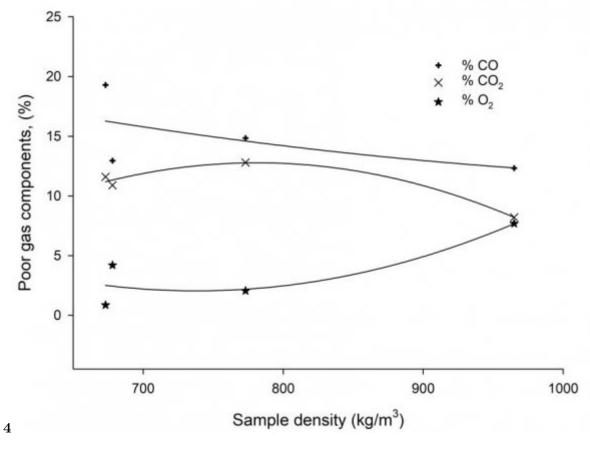


Figure 4: Figure 4 :

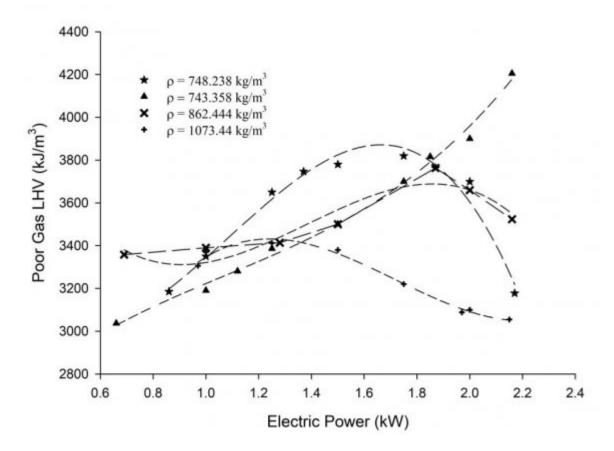


Figure 5:

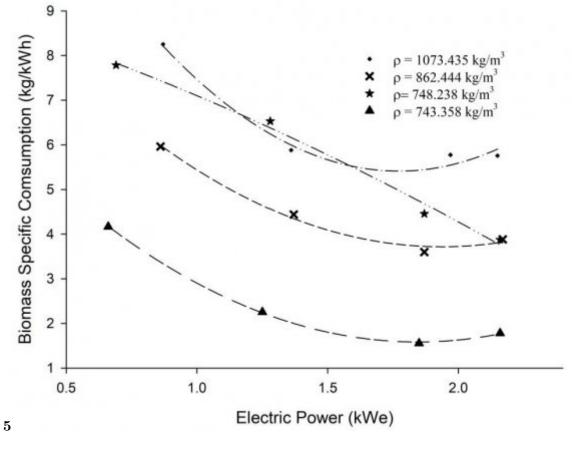


Figure 6: Figure 5 :

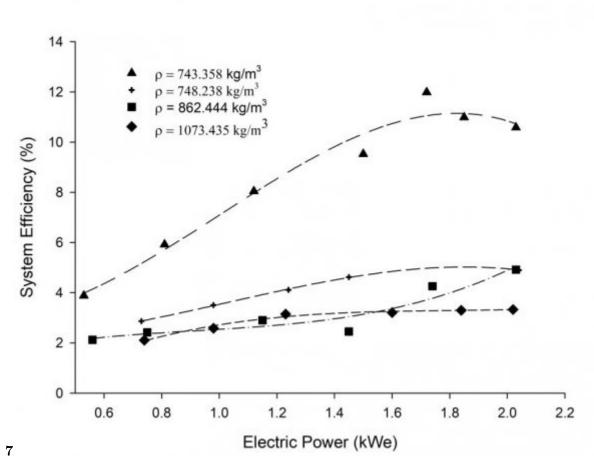


Figure 7: Figure 7:

1

	determined in triplicate		
Sample	1 Essay/ Moisture $Content(\%)$ 2	3 Avera	
1	$10.992 10.442 \qquad 9.042 10.159$		
2	8.280 10.149 9.304 9.244		
3	9.868 9.793 10.6700.110		
4	8.274 9.752 $9.5449.190$		
In Table 2, we can see that sample 1 properts a			

In Table 2, we can see that sample 1 presents a density 19.7 % larger than that of sample 3, which in turn has the second largest density among all the samples. Samples 2 and 4 have very similar density magnitudes. It should be noted that the average density of sample 1 is considerably higher as compared with the higher densities of different tropical species, see Reys et al. [53].

Figure 8: Table 1 :

3

Sample Volatile matter (%) Fixed carbon (%)		Ash $(\%)$	HHV	Moisture			
				(MJ/kg)	(%)		
1	91.470	4.390	4.140	15.780	11.090		
2	88.544	6.259	5.197	15.976	12.550		
3	96.215	2.186	1.599	15.760	11.730		
4	82.556	15.413	2.031	18.305	11.620		
c) Temperature Distribution Inside the Reactor							

c) Temperature Distribution Inside the Reactor

Table

Figure 9: Table 3 :

$\mathbf{4}$

Zone	Sample 1	Temperature (o	C) Sample 2 Sample 3	Sample 4			
Drying	40.5	52.5	61.5	45.6			
Pyrolysis	463.2	698.5	544.0	701.0			
Combustion	954.4	1028.0	1079.0	1162.0			
Reduction	860.0	844.0	952.7	1014.0			
d) Behavior of the gases CO, CO 2 and O 2 of the four							
biomass samples, with the engine running at							
maximum power							
In Figure							

Figure 10: Table 4 :

12 CONCLUSIONS

²⁶² .1 Funding sources

This research did not receive any specific grant from funding agencies in the public, commercial, or notfor-profit sectors.

²⁶⁵.2 Compliance with Ethical Standards:

- ²⁶⁶ The authors declare that they have no conflict of interest.
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