

Fabrication, Microstructure, Hardness and Magnetic Properties of (W:Ti)C-Ni Cemented Carbides using Atomized Ni Powder

Walid M. Daoush¹, Amal. A. Abd-Elghany², Mohamed A. Ghanem³ and Ahmed E. El-Nikhaily⁴

¹ Helwan University, Cairo, Egypt

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Abstract

(W:Ti)C cemented carbides with different Ni content fabricated by powder metallurgy technique were investigated. The Ni powder which used as a metal binder of the (W:Ti)C-Ni abrasive particles was fabricated by water atomization technique by means of induction furnace. A Ni powder of a cubical particle shape with average particle size of 2-20 μm was obtained. Six (W:Ti)C-Ni compositions with Ni contents of 5, 10, 15, 20, 25 and 30 wt.

Index terms— Fabrication, Microstructure, Hardness and Magnetic Properties of (W:Ti)C-Ni Cemented Carbides using Atomized Ni Powder

1 I. Introduction

emented carbides are used as oxygen-free ceramics in high temperature engineering applications due to its properties like, high melting temperature, hardness, elastic moduli, wear resistance, electric conductivity and high-temperature strength. Cemented carbides are also widely applied as a base of hard metals. Applications of cemented carbides include structural, heating and reflecting functions as well as tool materials in composition with other refractory compounds [1]. The compacted products of these compounds are made by powder metallurgy technology where sintering methods are of decisive importance.

In General; cemented carbides are composite materials, which consist of hard refractory carbides containing metals of the transition groups IV, V and VI (such as WC, TiC, TaC, NbC) embedded in a tough metal binder phase like nickel or cobalt which are by far the dominating binder metals employed due to its excellent wetting to WC and its good thermo-mechanical properties [2][3][4].

(W,Ti)C has a high melting point and high hardness than the commercial WC. In this regard, the transition metal carbide is primarily used in cutting tools and as an abrasive material as a single phase or in composite structures. In the case of cemented (W,Ti)C, Co or Ni is added as a binder for the formation of composite structures [5][6][7][8][9].

There are two basic ways of obtaining cemented carbide: through melt-solidification processing at about 2000-2500 °C or by powder metallurgy processing at a temperature range of 1350-1500 °C. Fabrications of WC as well as TiC in metalbase alloys have been studied from both theoretical and practical points of view. However, the production of (W:Ti)C in metal matrices has received little attention. The comparatively light TiC particles may float during the preparation by conventional melting and casting route. A large difference in density between TiC and the metal matrix melt-results segregation in metal ingot. The (W:Ti)C as reinforcements in nickel, cobalt and iron alloy melts may be more appropriate because its density (6.66 g/cm³ for (Ti 0.75 :W 0.25)C [10] and 9.1 g/cm³ for (Ti 0.5 :W 0.5)C [11]) is higher than that of TiC (4.25 g/cm³) and close to that of iron melt (7.8 g/cm³). The hardness of (Ti 1-x :W x)C (19-21 GPa) [12] is more or less the same as that of TiC (18-23 GPa) [13]. However, the fracture toughness of (Ti 1-x :W x)C (6.4-7.7 MPa m^{1/2}) [14] is higher than that of TiC (3.5-4.3 MPa m^{1/2}) [15,16] and this may result better mechanical properties of the (Ti 1-x :W x)C-reinforced composite compared to those of the TiC-reinforced composite.

The sintering step in the powder metallurgy process is the main determine step among the forming processes. It can be evaluated by measuring the mechanical properties like hardness and physical properties like density and

3 III. RESULTS AND DISCUSSION

46 magnetic properties of the sintered product. Magnetic data shows some interesting relations between the physical
47 properties, such as C relations between the physical properties, such as density, and the mechanical properties
48 such as hardness. There is interdependence between the mass saturation magnetization, M_s , and the coercive
49 force, H_c , both of which frequently occur in an inverse relation. M_s is mostly related to the amount of magnetic
50 material present, whereas H_c appears to be strongly influenced by the interaction between the particles and the
51 density or between the porosity and the grain size of the carbide phase [17,18].

52 In this work, cold compaction and vacuum sintering of (W:Ti)C cemented carbides samples with different
53 binder content of a homemade water atomized Ni powder were occurred. The densification and properties of the
54 prepared cemented carbide samples were characterized by microstructure investigations based on measurements
55 of magnetic and mechanical properties.

56 2 II. Experimental

57 Elemental powder of Ni was prepared by atomization technique. Nickel powder was prepared by induction melting
58 in graphite crucibles in air by superheated up to 1700°C and bottom pouring through a ceramic melt delivery
59 nozzle of 6 mm diameter into a confined water atomizer operating at a pressure of 20 MPa. The high-pressure
60 water jets were directed against the molten stream. The melt flow rate, estimated from the operating time and
61 weight of the atomized melt, was about 4 kg/min. The water flow rate, calculated from the water consumption
62 rate, was about 200 l/min. Table 1 list the atomization conditions adapted for Ni powder fabrication process.
63 The size distribution of the Ni powder particles was measured by conventional mechanical sieving, and sieved
64 powders with a specific size range of 20 μm and 1 μm were chosen for this investigation. The produced atomized
65 Ni and (W:Ti)C powders were used to prepare six samples with compositions of 5 wt.%, 10 wt.%, 15 wt.%, 20
66 wt.%, 25 wt.%, and 30 wt.% of a Ni binder mixed with (W:Ti)C by means of Agate mortar for 30 min. After
67 the different compositions were prepared, they under went cold compaction at 600 MPa in a uniaxial hydraulic
68 pressing machine, where they were compressed into a cylindrical shape.

69 The cold compacts were sintered in a vacuum furnace at 10⁻³ torr with graphite heating elements and at a
70 heating rate of 5 °C/min in accordance with the sintering cycle shown in Fig1. The samples were heated at 120
71 °C for 2 h to dry any moisture content, and the temperature was then raised to 750 °C for one hour to expel any
72 gases embedded in the pores. The temperature was raised again to 1450 °C for one hour to start the sintering
73 process. Finally, the furnace was turned off and the sintered compacts were cooled for 8 h by means of a water
74 cooling system.

75 The dimensions of the cold compacts were measured before and after sintering to calculate the green, the
76 sintered and the relative densities. The sintered samples were mounted and ground with 800, 1000, and 1200 grit
77 SiC paper, respectively, and then polished with 3 μm diamond paste. To investigate the microstructure of each
78 phase as well as the compositional analysis of the carbide and liquid phase binder, scanning electron microscope
79 (SEM, model: JEOL, JSM-5410) is used to take SEM micrographs and EDAX-SEM were used. The phases in
80 the specimens were analyzed by means of X-ray diffraction (XRD), for which purpose we used a Cu K α source
81 and a x-ray diffract meter of the model x , pert PRO PAN analytical with Cu k α radiation ($\lambda=0.15406\text{nm}$)
82 diffract meter. The magnetic properties of samples were measured using vibrating sample magnetometer (model
83 DEAS/FDD-2) in which the samples were vibrated at a constant frequency between a set of sense coils. As
84 the magnetic field is varied through a specified range up to 2 Tesla, the magnetic moment of the sample is
85 measured by the sense coils with a lock-in amplifier. The dependency between the magnetization and magnetic
86 field (hysteresis loop) for the prepared samples was measured. Because the saturation magnetization changes
87 with weight of the sample, the results were Global Journal of Researches in Engineering () Volume XVIII Issue I
88 Version I divided by sample's weight. The magnetization values were expressed using the magnetic moment per
89 gram (emu/g). The measured properties included saturation magnetization (M_s), coercivity (H_c) and remnant
90 magnetization (M_r). The hardness was measured with a Vickers hardness tester of the model Indentec 5030
91 SKG. The load was selected at 30 kgf. The test was repeated five times at different points in each sample, the
92 average being reported. The maximum attainable densification of the obtained (W:Ti)C-Ni cemented carbides
93 are due to the particles rearrangement is influenced by different parameters, such as the amount of liquid present,
94 the particle size, the contact angle, and the solubility of the solid in the liquid. Fig. 5 shows the results of the
95 micro structural investigation with respect to the different metal binder content for (W:Ti)C-Ni. We can see,
96 firstly, that the porosity of the materials decreases as the Ni content increases. However, when the Ni content is
97 high, the Ni metal binder prevents the coalescence of the carbide particles, producing a more uniform and finer
98 grain structure. In other words, all the carbide grains are separated by a layer of the nickel metal binder and
99 consequently produce a normal grain growth [20][21][22][23].

100 3 III. Results and Discussion

101 The results of the high resolution SEM and the composition analysis (EDAX), as shown in Fig. 6, clearly reveal
102 that the (W:Ti)C particles have a rounded morphology with a carbide core/rim structure. This morphology is
103 formed when the carbide particles are dissolved in the Ni liquid and re-precipitated on the large carbide grains,
104 where they cause grain coarsening, called Ostwald ripening [24,25].

4 Intensity

(a.u.) Fig. 8 shows the results of the measured hardness of the obtained (W:Ti)C-Ni cemented carbides indicated that the exact character of the microstructure has a critical influence on the resultant hardness. The insitu hardness of nickel, namely 440, is much higher than the typical values of 240 for bulk nickel. This difference is explained by the solid solution of W, Ti and C in the liquid phase binder and by the complex stress state resulting from the different thermal expansion coefficients of nickel and W, Ti and. The compositional analysis confirms that the hardening effect of nickel is due to the solid solution of Ti in the Ni binder. Moreover, due to the solubility of WC in the TiC forming (W:Ti)C phase, the hardness value of the (W:Ti) C phase is 2530, which is higher than the micro-hardness value of the WC (1300) and lower than that of the TiC (3200). Fig. 9 also shows increasing in the hardness by increase the metal binder content. This result can be discussed due to the densification effect. By increasing the metal binder content the porosity was decreased, the densification was increased and as a result the hardness increased and by increasing the Ni binder content which has lower hardness than the carbide phase [26]. that are separated or combined in a solution; and, third, the presence of carbide grains, which act as magnetic voids where opposing magnetic fields can occur and in turn reduce the saturation magnetization values [28:29]. Fig. 12 illustrates the influence of the Ni binder on the coercivity (H_c). First, the coercivity decreases with increasing the metal binder content. Second, the sinter ability increases and the porosity decreases, thereby decreases the particle-particle interactions and decreasing the coercivity of the materials because of the porosity values of the (W:Ti)C-Ni, and by decreasing the porosity the particle-particle interaction was decreased and the coercivity decreased [30][31][32][33].

5 2? (degree) (W, Ti) C Ni

6 IV. Conclusion

The microstructure, hardness and magnetic properties of (W,Ti)C-Ni cemented carbides were investigated. The (W:Ti)C cemented carbides were consolidated with a water atomized Ni as the liquid phase binder by using the vacuum liquid phase sintering at 1450 o C. The major results are summarized as follows:

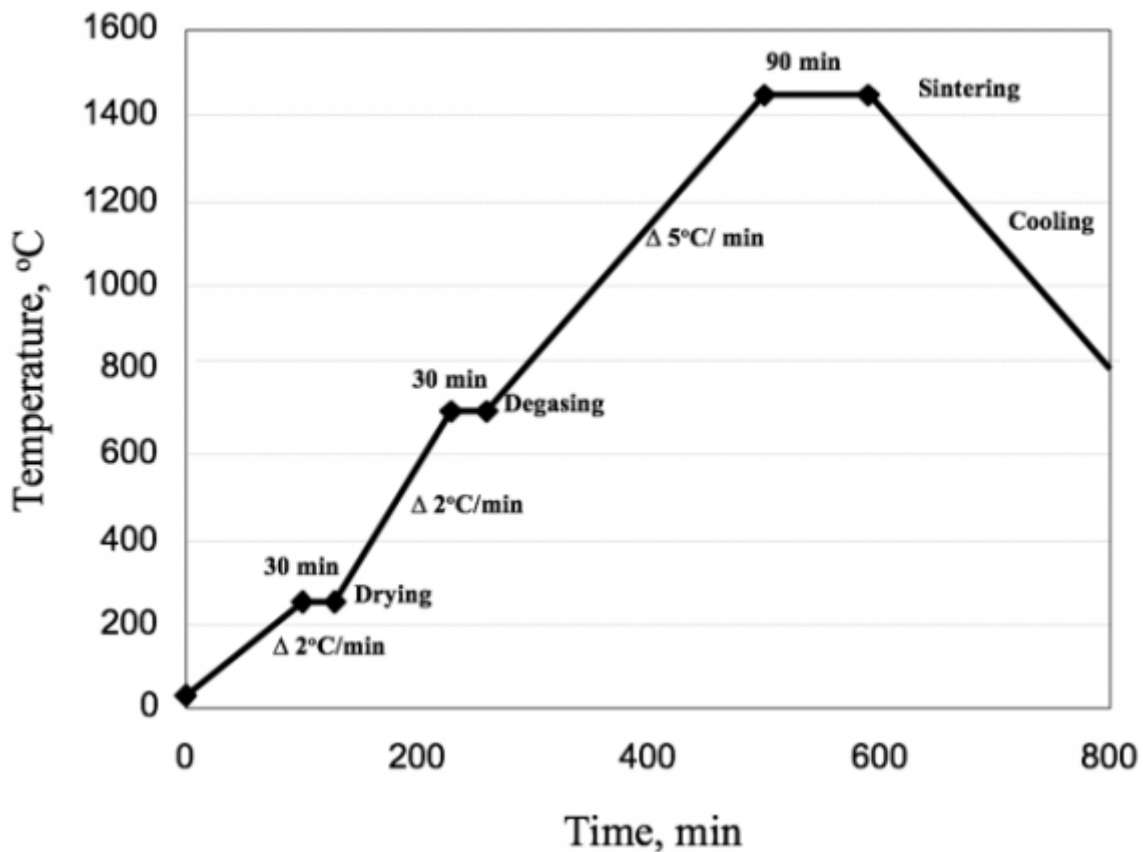
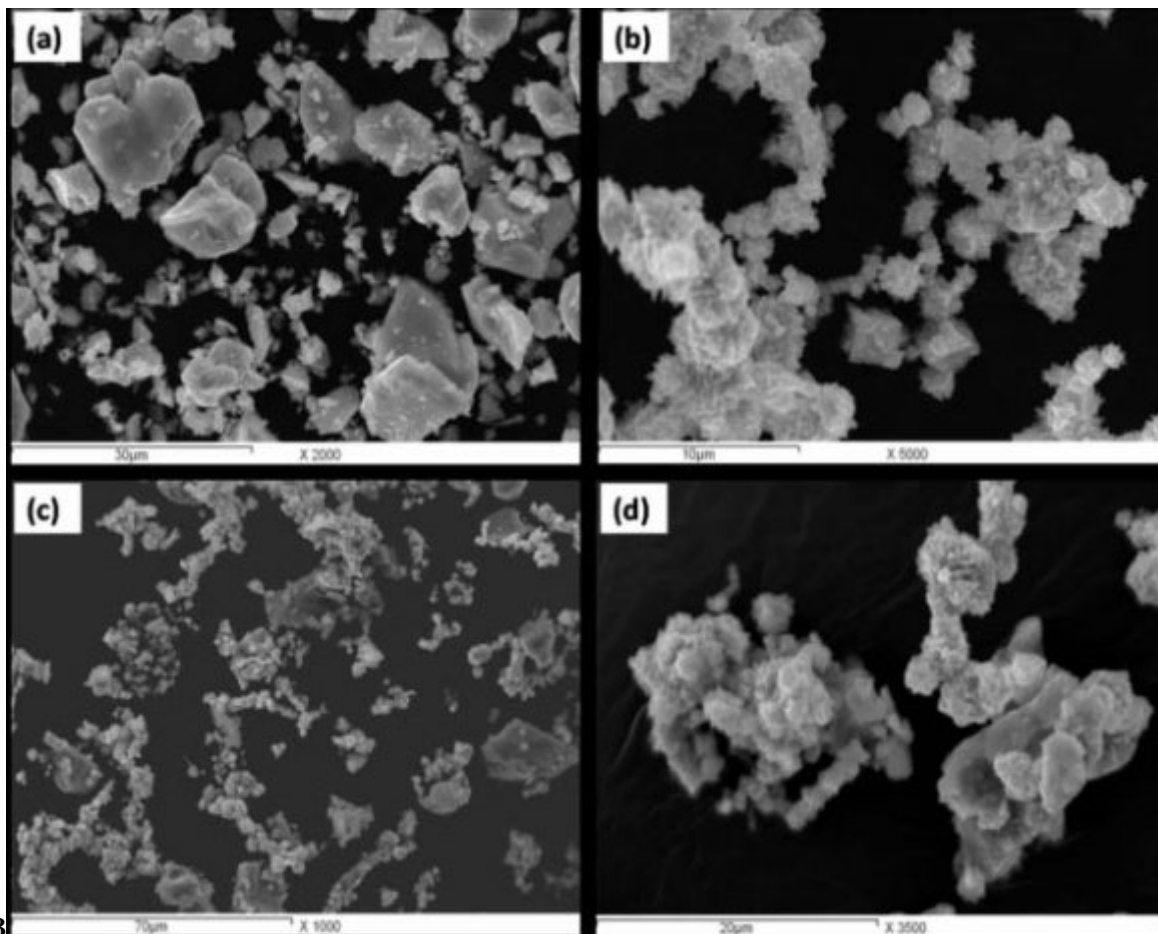
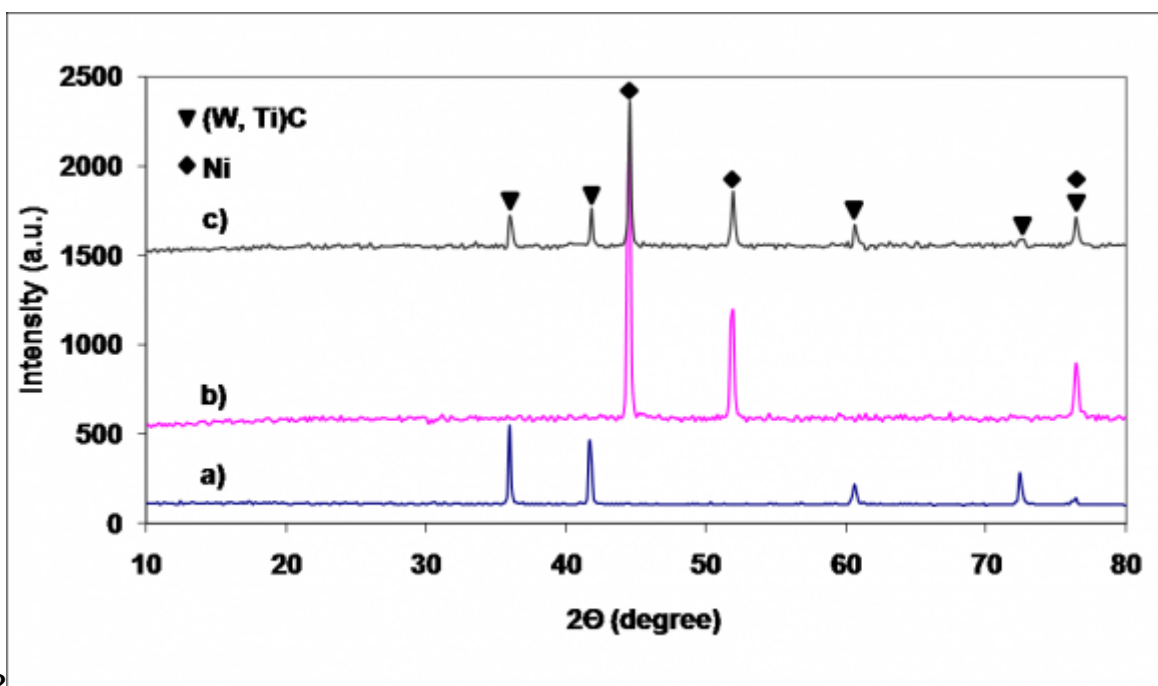


Figure 1: Figure 2 :



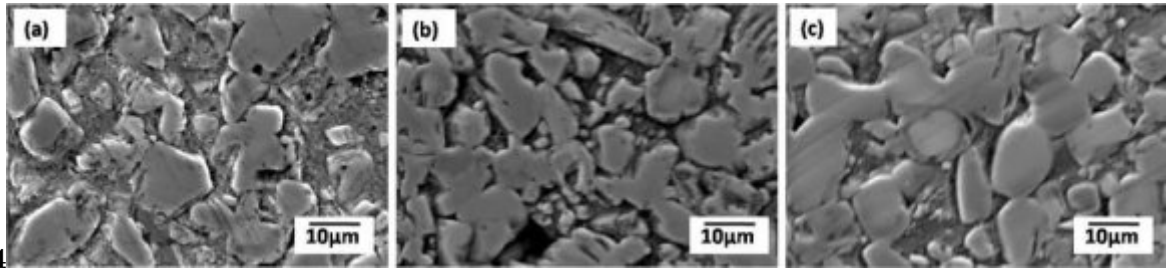
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Figure 2: Figure 3 :



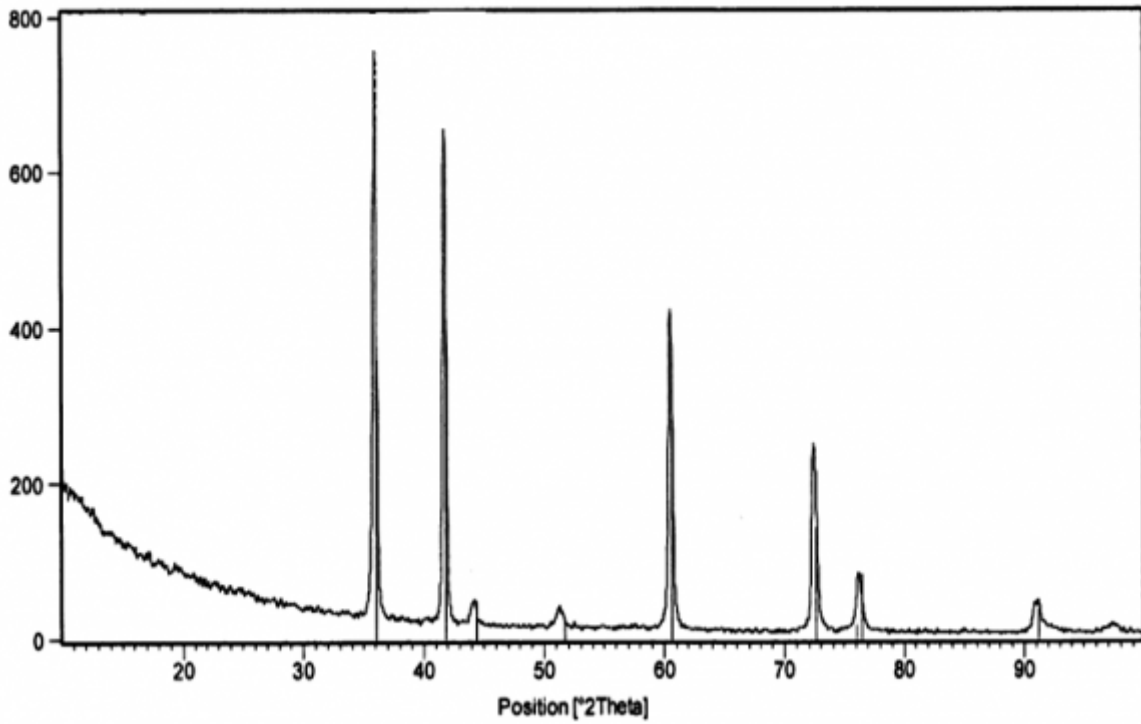
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Figure 3: Fig. 2 (



4

Figure 4: Figure 4 :



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Figure 5: Figure 5 :

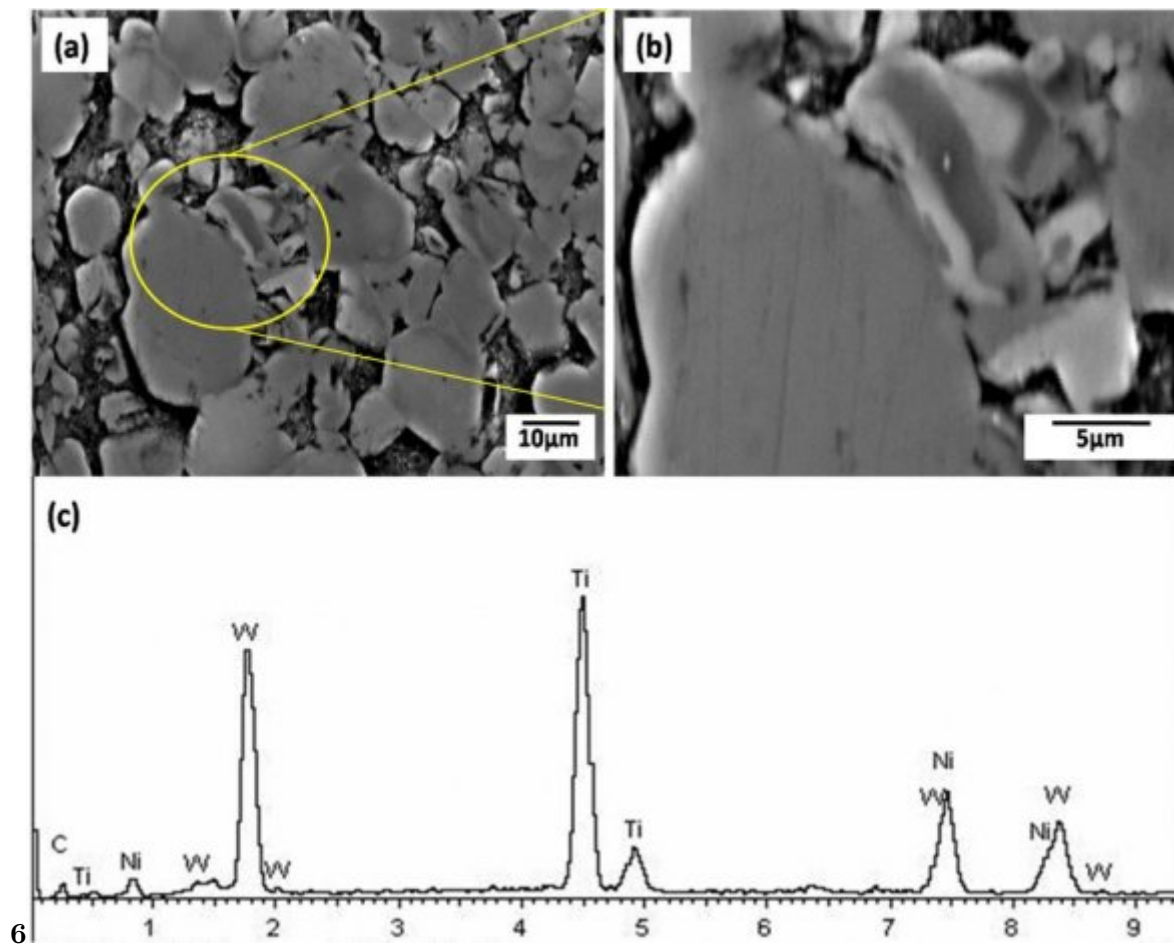


Figure 6: Figure 6 :

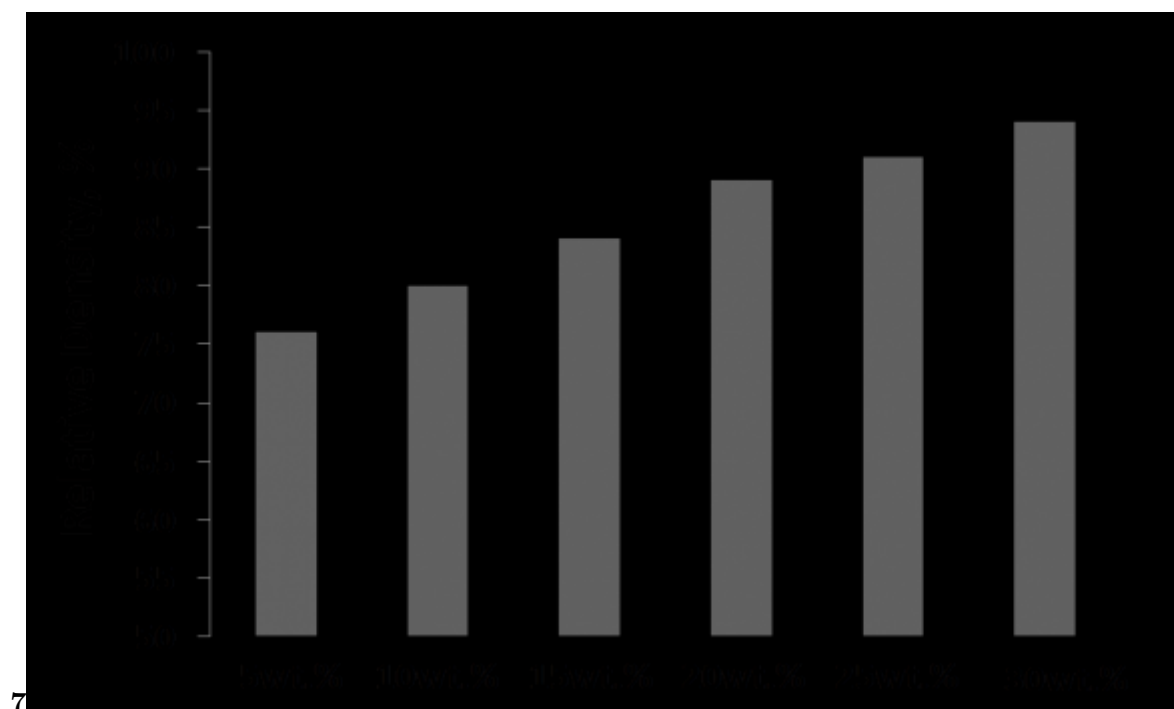
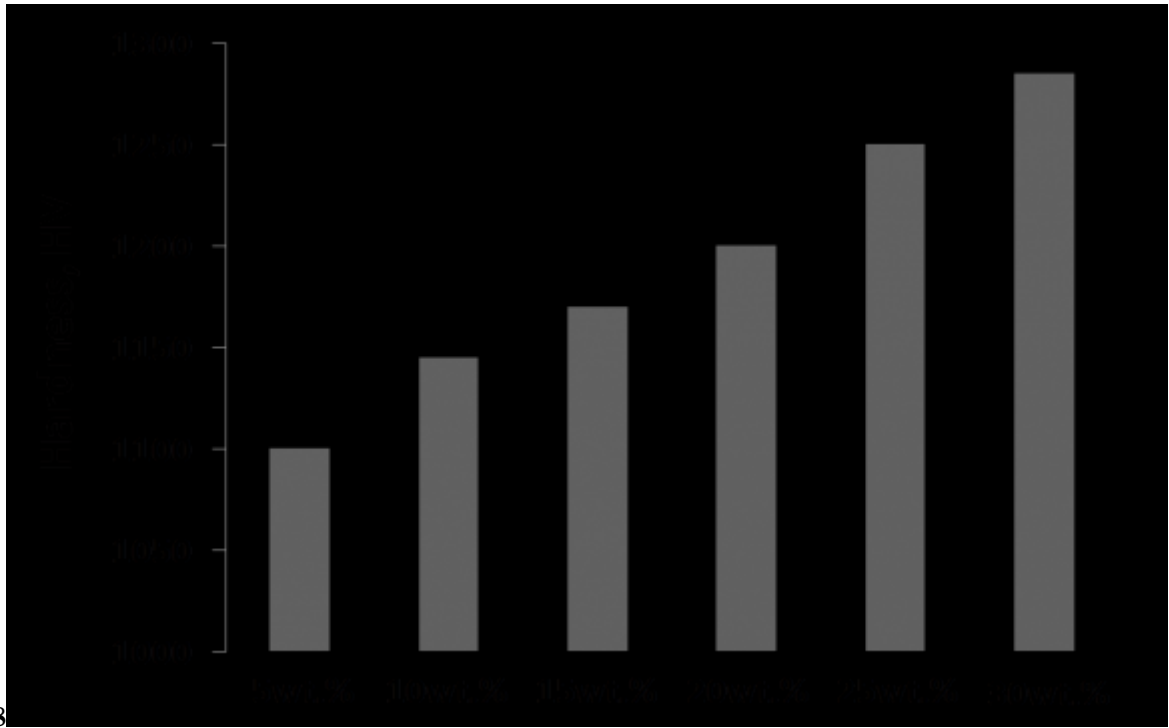
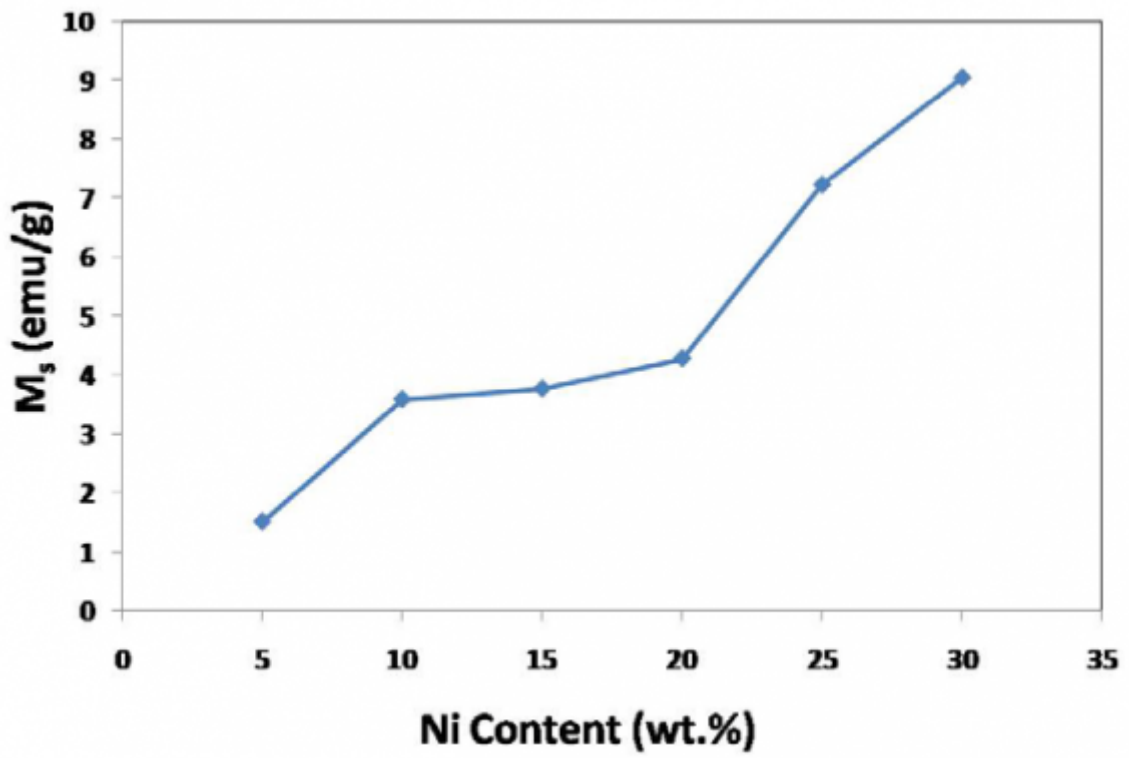


Figure 7: Figure 7 :



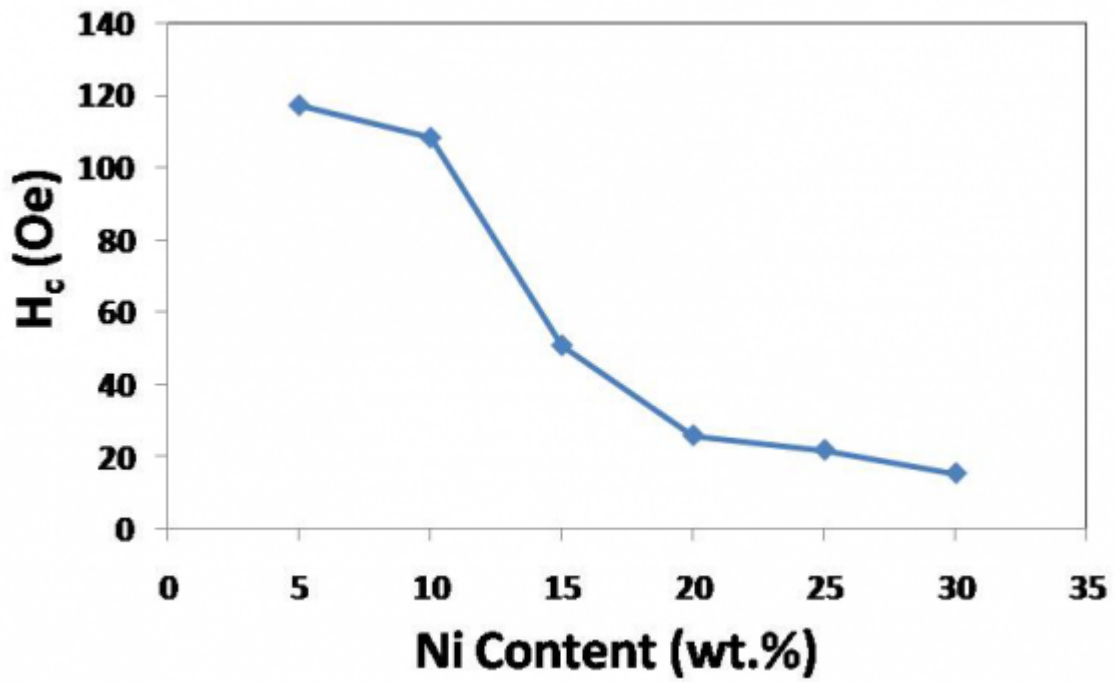
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Figure 8: Figure 8 :



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Figure 9: Figure 9 :



10

Figure 10: Figure 10 :

1

Parameter	Condition
Pouring temperature, o C	1700
Nozzle angle	35 o
Nozzle diameter, mm	6
Number of water jets	4
Molten stream flow rate, kg/min.	4
Water pressure, MPa	20
Water flow rate, l/min.	200
Water velocity, m/s	90

Figure 11: Table 1 :

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6 IV. CONCLUSION

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