

Performance Simulation of Two-Bed Silica Gel-Water Adsorption Chillers

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Abstract

This paper presents a transient model of a two bed silica gel - water solar adsorption cooling system. This program is then utilized to simulate the performance of a sample solar adsorption cooling system used for cooling a room that comprises an area of 9 m² located in Nancy city in France. The system has been simulated with typical weather data of solar radiation and ambient temperatures of France. The results include effects of the hot water temperature, cooling water temperature and chiller water temperature and cycle time on COP, refrigeration capacity and cycled mass are studied in order to determine their optimum values able to maximize the overall performance of the system under analysis for its adaptation to the Tunisian climate.

Index terms— solar adsorption chiller, silica gel-water, simulation, performance.

1 I. Introduction

With the increasing economic development and environment protection, adsorption refrigeration technology as the green refrigeration method has received more and more attention in recent years because it can save energy and is environmentally friendly. Adsorption refrigeration can be driven by lowgrade heat source, such as waste heat from the process industry and solar energy.

The adsorption cooling and heat pump systems could utilize low temperature waste heat or renewable energy sources. The working pairs of adsorption cooling and heat pump are mainly dominated by activated carbon/ammonia, activated carbon/methanol and activated carbon fiber (ACF)/ammonia, silica gel/water and zeolite/water pairs.

In this context, silica gel-water was selected as the adsorbent-adsorbate pair. Compared with other adsorbents, silica gel can be regenerated at a relatively low temperature (below 100°C, and typically about 85°C). It also has a large uptake capacity for water which has a high latent heat of evaporation; up to 40% of its dry mass. A silica gel-water adsorption chiller is able to make use of industrial waste heat to effect useful cooling.

Many researchers evaluated the performance of adsorption cooling and heat pump systems based on working pairs, system design and methodology. A transient simulation model for adsorption cooling system using silica gel/water pair powered by renewable energy was investigated by a number of researchers [1][2][3][4][5][6][7]. Restuccia et al. [8] reported an experimental and numerical study of a lab-scale adsorption chiller using the macroporous silica gel impregnated with CaCl₂ as the adsorbent. At a 90-95°C heat source, the authors showed that the measured COP values were up to 0.6. With the aim of improving silica gel-water adsorption chillers design with two adsorption/desorption chambers, Liu et al. [9] demonstrated that with the new chiller, a COP of about 0.5 is reached. In the same way, Núñez et al. [10] presented the development of a prototype of a small adsorption heat pump using silica gel-water pair. The purpose of minimizing primer energy consumption is achieved. In fact, for air-conditioning of 12-15°C, a cooling COP of 0.5 is found. Wang et al. [11] built and tested a novel silica gel-water adsorption chiller. For a hot water temperature of 84.8°C, a cooling water temperature of 30.6°C and a chilled water outlet temperature of 11.7°C, the measured COP is about 0.38. The authors proved

2 II. EXPERIMENTAL DEVICE

44 that the application of this adsorption chiller is successful especially for low grade heat source. Xia et al. [12]
45 presented an improved two bed silica gel-water adsorption chiller. The improved chiller is composed of three
46 vacuum chambers: two adsorption/desorption vacuum chambers and one heat pipe working vacuum chamber.
47 A heat pipe is used to combine the evaporators of the two adsorption/ desorption units. An improvement of at
48 least 12% for the COP was reached compared to the formers chillers. Hen et al. [13] investigated an improved
49 compact silica gel-water adsorption chiller without vacuum valves. To improve the performance of the chiller, a
50 heat and mass recovery process is carried out. The COP is measured about 0.49. Liu et al. [14] developed a
51 new adsorption water chiller without refrigerant valves. The working pair is silica gel-water with mass recovery
52 process. The COP range was 0.2-0.42 depending on the operating conditions.

53 Saha et al. [15] proposed a new two-stage nonregenerative adsorption chiller design and experimental prototype
54 silica gel-water adsorption chiller. To exploit solar/waste heat of temperatures below 70°C, staged regeneration is
55 necessary. The two-stage cycle can be operated effectively with 55°C solar/waste heat in combination with a 30°C
56 coolant temperature. He et al. [16] carried out a novel two stages adsorption chiller with different adsorbents such
57 as Zeolite and activated carbon. The two-stage cycle can be operated effectively with a generator temperature
58 of 45-50°C.

59 A two stage activated carbon cycle using R134a and R507A refrigerants in the two stages was investigated by
60 Habib et al. [17]. The evaporator of the R134a cycle was connected to the condenser of the R507a system. The
61 performance in this cycle was comparatively low, achieving COPs of only 0.04-0.1.

62 A novel three-bed, two-evaporator system was proposed and modeled by Miyazaki et al. [18]. The dual
63 evaporator allows two beds to be adsorbing simultaneously, while a third is desorbing. A bed is connected to a
64 low pressure evaporator and then when reaching near saturation conditions for that bed, it is connected to a high
65 pressure evaporator and adsorption continues. COP for this system design increased by 70%, while SCP increased
66 by 50% for this system design compared to a standard adsorption chiller working at the same conditions.

67 Several configurations were investigated by Li et al. [19], including a bed-to-bed re-adsorption process. It was
68 found that using a bed-to-bed system improved the cooling capacity of the system by delivering cooling at both
69 the evaporator and first adsorbent bed, although at different cooling output temperatures. The bed-to-bed design
70 was also made adaptable so that the process could incorporate internal heat recovery, depending on the desired
71 output, or operate as a conventional system. The COP doubled when operating in bed-to-bed mode compared to
72 conventional operation. K. Habib et al. [20] presents the theoretical analysis of the performance of solar powered
73 combined adsorption refrigeration cycles that has been designed for Singapore and Malaysia and similar tropical
74 regions using evacuated tube solar collectors. This novel cycle amalgamates the activated carbon (AC)-R507A
75 as the bottoming cycle and activated carbon-R134a cycle as the topping cycle and deliver refrigeration load as
76 low as -10 °C at the bottoming cycle. A simulation program has been developed for modeling and performance
77 evaluation for the solar driven combined adsorption refrigeration cycle using the meteorological data of Singapore
78 and Malaysia. The results show that the combined cycle is in phase with the weather. The optimum cooling
79 capacity, coefficient of performance (COP) and chiller efficiency are calculated in terms of cycle time, switching
80 time, regeneration and brine inlet temperatures.

81 A. Sadeghlu et al. [21], divided combined ADRS into four types based on different arrangements of two working
82 pairs, Zeolite 13x/CaCl₂-water and Silica gel (RD type)-water, to analyze the performance of combined ADRS.
83 After validating mathematical models with available experimental data, ADRS is simulated by using Simulink-
84 Matlab software to achieve optimum times for various processes. The results of simulation show that the cooling
85 capacity of the system with Zeolite 13x/CaCl₂-water is more than the other types. The results have shown
86 that the arrangement of adsorbents affects cooling capacity of combined ADRS significantly. In Type A, Zeolite
87 13x/CaCl₂-water has been used as an adsorbent for both top and bottom cycles. This type not only has more
88 cooling capacity than the other types, but also the effect of hot water temperature on cooling capacity of this
89 type is less than the others. Furthermore, a sensitivity analysis has been done to determine the importance of
90 each parameter on ADRS system because the cooling capacity and the COP are influenced by many constant
91 parameters.

92 The objective of this paper is the development of a global simulation model flexible in changing operating
93 conditions using Simulink. The optimization tools are used to enable selecting the optimum operating conditions
94 corresponding to the best performance in order to adapt this machine in to Tunisian climate that having a cooling
95 temperature up to 40°C and heating temperature up to 85°C.

96 2 II. Experimental Device

97 Figure 1 illustrates the experimental unit, driven by solar energy, provided from solar collectors, and the fuel
98 source such as the natural gas. This platform combine cogeneration (by the production of electricity and heating),
99 solar cooling, and sustainable construction (wood structure). Two similar adjacent chambers, with opposite
100 comfort demand, are the users of heating and refrigeration.

101 The major components contained in the platform and ENERBAT which are included in the experiments
102 carried out are: Solar panel: On the roof, a solar field with 16 solar collectors, 2.4 m² each is installed. The
103 collector characteristics are given in the following:

104 Hot water tank thermal stratification: The heat provided by the solar panel or by the co-generator is stored

105 in the hot water cylinder, to thermal lamination, with a capacity of 1500 liters. The hot water fed from the tank
 106 to the adsorption refrigerating machine.

107 **3 Dry cooler:**

108 The dry cooler constitutes the cooling circuit of the machine adsorption Two-room climate: it consists of two
 109 rooms a warm room and a hot room represents the test cell.

110 Co-generator: The used co-generator is an internal combustion engine coupled with electric generator which
 111 recovers more than 90 % of heat from coolant, lubricant, and exhaust gas. Thus, it is used as mean of producing
 112 electricity (220 V, 50 Hz) and heat (hot water at 85°C). Its electrical and thermal efficiencies are approximately
 113 25 % and 65 %, respectively. Data acquisition: The data are acquired and manipulated as two dimensional
 114 graphs and tabulated. Instrumentation also allows regulation of the trigeneration unit.

115 In this paper we focus only on the refrigeration machine for which we developed a simulation model that is
 116 confronted to experimental measurements. In order to develop a mathematical model, a number of assumptions
 117 are required. The temperature, pressure and the amount of water vapor adsorbed are uniform throughout
 118 the adsorber beds. There is no external heat loss to the environment as all the beds are well-insulated. The
 119 condensate can flow into the evaporator easily. All desorbed water vapor from the desorber will flow
 120 into the condenser immediately and the condensate will flow into the evaporator directly. The condensate will
 121 evaporate instantaneously in the evaporator and will be adsorbed in the adsorber immediately. The adsorbed
 122 phase is considered as a liquid and the adsorbate gas is assumed to be an ideal gas.

123 The thermal resistance between the metal tube and the adsorbent bed is neglected. Flow resistance arising
 124 from the water flowing in the pipeline is neglected. The properties of the fluid, the metal tube and adsorbate
 125 vapor are constant.

126 According to these assumptions, the dynamic behavior of heat and mass transfer inside different components
 127 of the adsorption chiller can be written as shown below.

128 **4 b) Rate of Adsorption/desorption**

129 The rate of adsorption or desorption is calculated by the linear driving force kinetic equation, The coefficients of
 130 LDF equation for silica gel/water are determined by Chihara and Suzuki [22] and are given in the below:

$$131 = K_s (w^* - w) \text{ (kg/kg.s)} \quad (1)$$

132 The effective mass transfer coefficient inside the pores k_s is given by:

$$133 k_s = F_0 D_s R_p^2 (s-1) \quad (2)$$

134 The effective diffusivity is defined as follows:

$$135 D_s = D_{s0} \exp(-E_a/RT) \quad (m^2/s) \quad (3)$$

136 Where $P_s(T_w)$ and $P_s(T_s)$ are respectively the corresponding saturated vapor pressures of the refrigerant
 at temperatures T_r (water vapor) and T_a (adsorbent). P_s for water vapor is estimated using the following
 equation:

$$137 P_{sat}(T) = 133,32 \exp(18,3 - 3820/T + 46,1) \quad (5)$$

137 **5 c) Energy balance of adsorber**

138 The adsorption energy balance is described by:

$$139 (m_{ad} c_{ad} + m_{ac} c_a + m_{wcp} r) dT_{ad} dt = m_a ?H_{ads} dw dt + m_a c_{p,r,v} dw dt (T_{ev} - T_{ad}) + m_{f,ad} c_{p,f} (T_{f,in} - T_{f,out}) \text{ (kW)} \quad (6)$$

140 The outlet temperature of cooling water can be expressed as T_{ad}

142 **6 d) Energy balance of desorber**

143 The desorption energy balance is described by:

$$144 (m_{de} c_{de} + m_{dc} c_a + m_{wcp} r) dT_{de} dt = m_a ?H_{ads} dw dt + m_{f,de} c_{p,f} (T_{f,in} - T_{f,out}) \text{ (kW)} \quad (8)$$

145 The outlet temperature of hot water can be expressed as $T_{de,out} = T_{de} + (T_{de,in} - T_{de}) \exp(-U_{de} A_{de} m_{f,de} / (c_{p,f,de})) \quad (9)$

147 **7 e) Energy balance of condenser**

148 The condenser energy balance equation can be written as

$$149 (m_{r,cd} c_{p,r} + m_{cd} c_{cd}) dT_{cd} dt = m_a dw_{des} dt L_v + m_a c_{p,r,v} dw_{des} dt (T_{de} - T_{cd}) + m_{f,cd} c_{p,f} (T_{f,in} - T_{f,out}) \text{ (kW)} \quad (10)$$

150 The outlet temperature of cooling water can be expressed as

$$151 T_{cd,out} = T_{cd} + (T_{cd,in} - T_{cd}) \exp(-U_{cd} A_{cd} m_{f,cd} / (c_{p,f,cd})) \quad (11)$$

152 **8 f) Energy balance of evaporator**

153 The energy balance in the evaporator is expressed as

$$154 (m_{ev} c_{ev} + m_{r,ev} c_{p,r}) dT_{ev} dt = m_a dw_{ads} dt L_v + m_a c_{p,r,v} dw_{ads} dt (T_{cd} - T_{ev}) + m_{f,ev} c_{p,f} (T_{f,in} - T_{f,out}) \text{ (kW)} \quad (12)$$

155 The outlet temperature of chilled water can be written as

$$156 T_{ev,out} = T_{ev} + (T_{ev,in} - T_{ev}) \exp(-U_{ev} A_{ev} m_{f,ev} / (c_{p,f,ev})) \quad (13)$$

157 **9 g) Mass balance in the evaporator**

158 The mass balance for the refrigerant can be expressed by neglecting the gas phase as:

159 Where, m_a is the adsorbent mass.

160 10 h) System performance equations

161 The COP value is defined by the following equation: $COP = \frac{Q_{ev}}{Q_{de}}$ (15)

162 The cooling capacity of the system is expressed by: $Q_{ev} = \dot{m} c_p (T_{ev,in} - T_{ev,out}) \frac{1}{\Delta t_{cycle}}$ (16)

163
164 Where: 2 shows the experimental and numerical temperature profiles of the hot, cooling and chilled water.
165 After about 7mn, the hot water outlet temperature approaches to the inlet temperature, thus the heat consumed
166 by the desorber after this point, will be quite small. But the difference between outlet and inlet temperature
167 for the cooling water 1.8°C after cooling the adsorber for 7mn which shows that adsorber is sufficiently cooled
168 down and the adsorption ability remains strong until the end of adsorption phase. Therefore the cycle time is
169 taken as 14mn. It is worthy of note that the difference between outlet and inlet temperature of hot water after
170 heating the desorber for 7mn is 3°C. It is also observed that the outlet temperature of chilled water reaches its
171 minimum after each bed is heated/cooled for 50s. At this point the cooling power is at its maximum and the
172 outlet temperature of chilled water is 11.8°C. The switching time is taken as 40s. $Q_{de} = \dot{m} c_p \Delta T$,

173 11 b) Parametric Study of the adsorption machine

174 Cooling / Heating/chilled water inlet temperature influences adsorption chiller performance. Lowering cooling
175 water inlet temperature not only increases cooling capacity, but also enhances adsorption chiller COP, due to the
176 significant increase in adsorption rate. Increasing heating water temperature also enhances chiller cooling capacity
177 due to enhancing desorption rate that generate the adsorbed refrigerant prior to the evaporation/adsorption mode.
178 However, it negatively influences the chiller COP depending on the cooling water inlet temperature.

179 12 ? Effect of hot water inlet temperature

180 Figure 3 presents the change in chiller cooling capacity (SCP) and COP versus hot water inlet temperature at
181 various cooling water inlet temperatures. Other operating conditions (cycle time, chilled water inlet temperature
182 and secondary fluid flow rate) remain constant at their design values. As the hot water inlet temperature
183 increases the chiller cooling capacity increases for all cooling water inlet temperatures. As for COP, with hot
184 water temperature variation from 55 to 95 °C, COP increases. Because a higher hot water temperature causes a
185 higher heating power as well as a higher refrigerating capacity. For temperatures below 85°C, remained relatively
186 constant with the increase in the generation temperature, this is due to the insufficient refrigerant circulation
187 required to generate the cooling power.

188 It is clear that the sorption process is much faster for the highest temperatures. This means that the increase
189 of hot water inlet temperature allows an increase in the rate of desorption and thereafter a faster heat transfer
190 that generates the refrigerant adsorbed before the evaporation / adsorption phase.

191 Lowering the cooling water temperature increases the specific cooling capacity and coefficient of performance,
192 because the condensation is faster for lower condenser cooling water temperatures, also because of the increase in
193 adsorption rate. Figure 4 shows the change in the outlet chilled temperature versus hot water inlet temperature
194 for variable cooling water inlet temperature, there is a slight variation of the evaporator outlet temperature that
195 decreases with an increasing the heating water temperature. The efficiency shows the ratio between the actual
196 coefficient of performance and the Carnot cycle coefficient of performance ideal inverse (Figure 5).

197 The Carnot coefficient of performance is calculated by the following relation: $COP_{carnot} = \frac{T_{ad}}{T_{de} - T_{ad}}$ (19)

198
199 The adsorption efficiency of the machine is determined by: $\eta = \frac{COP}{COP_{carnot}}$ (20) Efficiency increases
200 with increasing hot water inlet temperature and lowering the cooling water inlet temperature. ? Effect of chilled
201 water inlet temperature In this part, the hot water inlet temperature is set at 85 ° C and the cooling water inlet
202 temperature is 40 ° C (as Tunisian conditions), we will vary the chilled water inlet temperature and see the effect
203 on the performance of the adsorption chiller.

204 Figure 6 shows the change in COP and SCP versus the inlet evaporator temperature, it is noted that for a
205 variation of the latter to 20 ° C a variation of COP and SCP respectively 0.2 and 0.481 kW / kg; thus increasing
206 the evaporator inlet temperature increases evaporation rates and then increase the cold production thus increasing
207 system performance.

208 13 ? Effect of cycle time

209 The refrigeration capacity and COP variations with the cycle time are shown in Figure 8, the Operating conditions
210 are setting en table 1. The COP increases uniformly with extension of the cycle time under a driving heat source
211 of 85°C. This is because a longer cycle time causes much lower consumption of driving heat, the maximum COP
212 can be obtained at maximum adsorption / desorption time, which correspond to the minimum heating capacity
213 and maximum adsorbed refrigerant amount. Based on the aforementioned results, the aim is to have a short
214 cycle time with a reasonable performance, so the optimal time 1240s cycle can be a tool to optimize adsorption
215 system. Figures 9 present the variation of COP and SCP according to the hot water inlet temperature. Indeed,

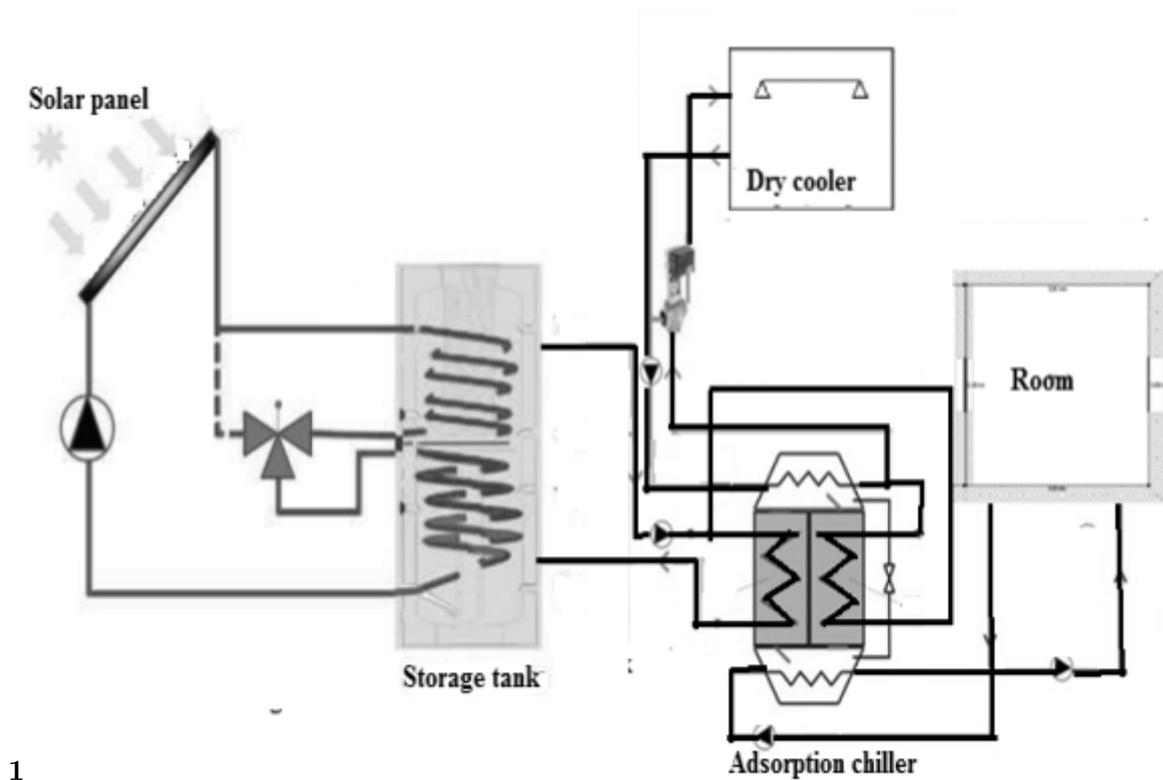
216 water vapor is desorbed rapidly to a higher regeneration temperature to desorb most of the water vapor to be
217 adsorbed in the next adsorption process.

218 Curves COP and the SCP for different adsorbents; silica gel, activated carbon and adsorbent composite (silica
219 activated carbon/ CaCl_2)/eau, shows that for adsorbent composite, the COP and SCP is greater.

220 14 V. Conclusion

221 This work presents a solar adsorption refrigeration system using silica gel / water pairs. We have developed a
222 numerical model for simulating the heat and mass transfer of the adsorption and regeneration processes in the
223 two beds. This allowed us to study the influence of the regeneration, cooling and evaporator inlet temperature
224 on the performance of the machine. The results show that the study parameters have a great impact on system
225 performance for its adaptation to the Tunisian climate. It is preferable to work with a high regeneration and
226 evaporation temperature where the coefficient of performance reaches its maximum value and a lower temperature
227 at the cooling water of condenser and adsorber. The adaptation of chiller to the Tunisian climate was made. for
228 a hot water inlet temperature of 85°C and a cooling water inlet temperature of 40°C we had a $\text{COP}=0,3$ and an
229 $\text{SCP}= 57 \text{ W/kg}$.

230 15 Global



1

Figure 1: Fig. 1 :

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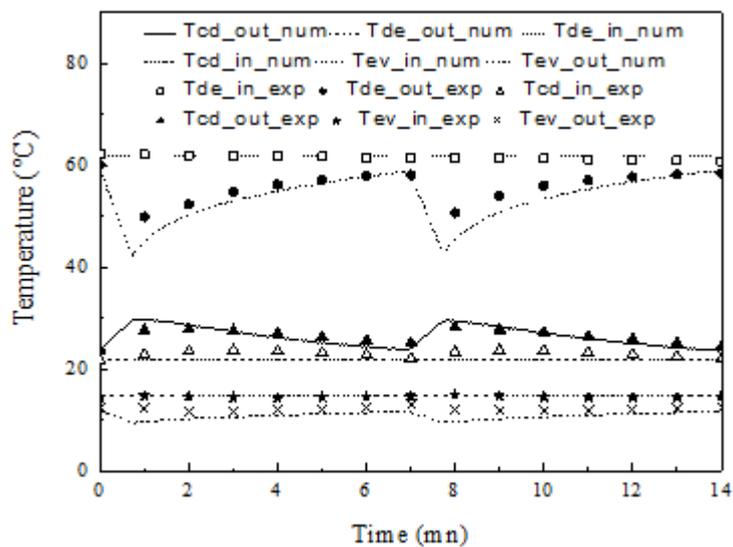
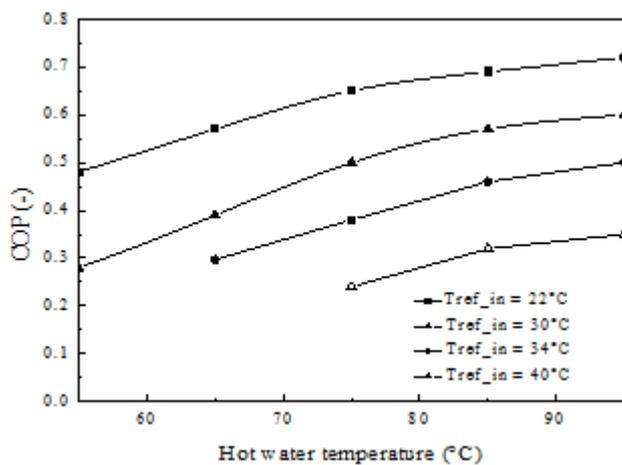
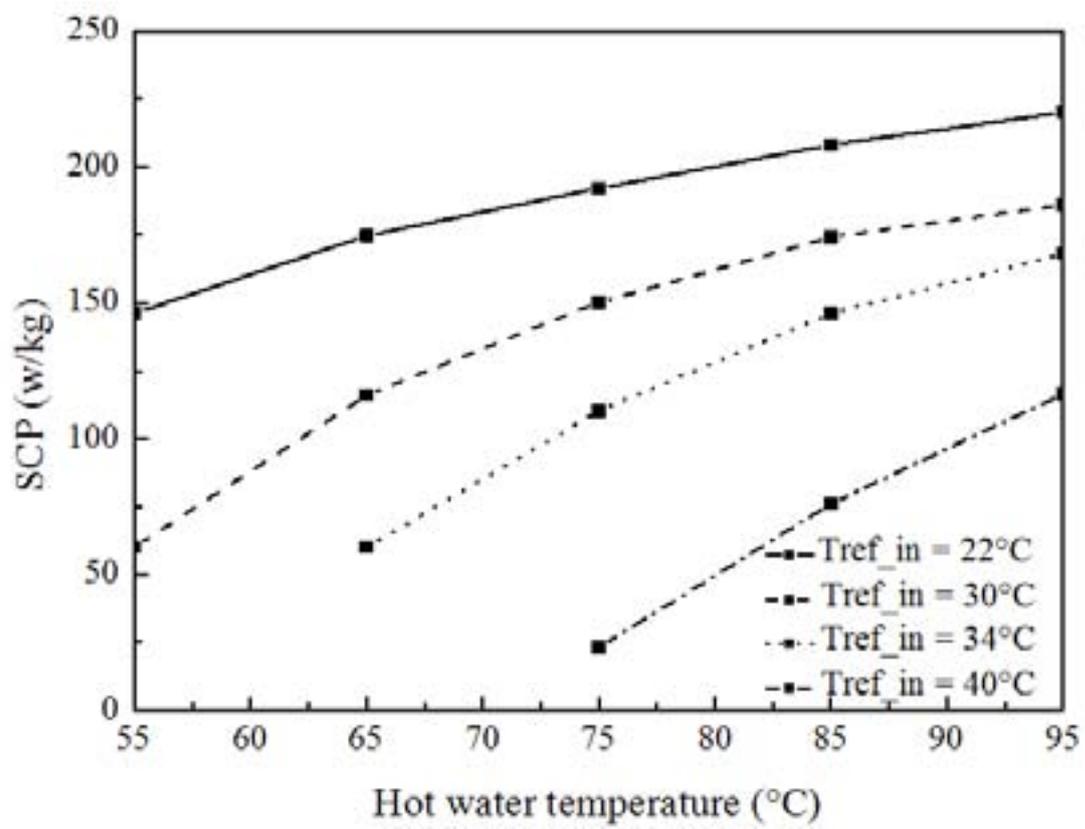


Figure 2:



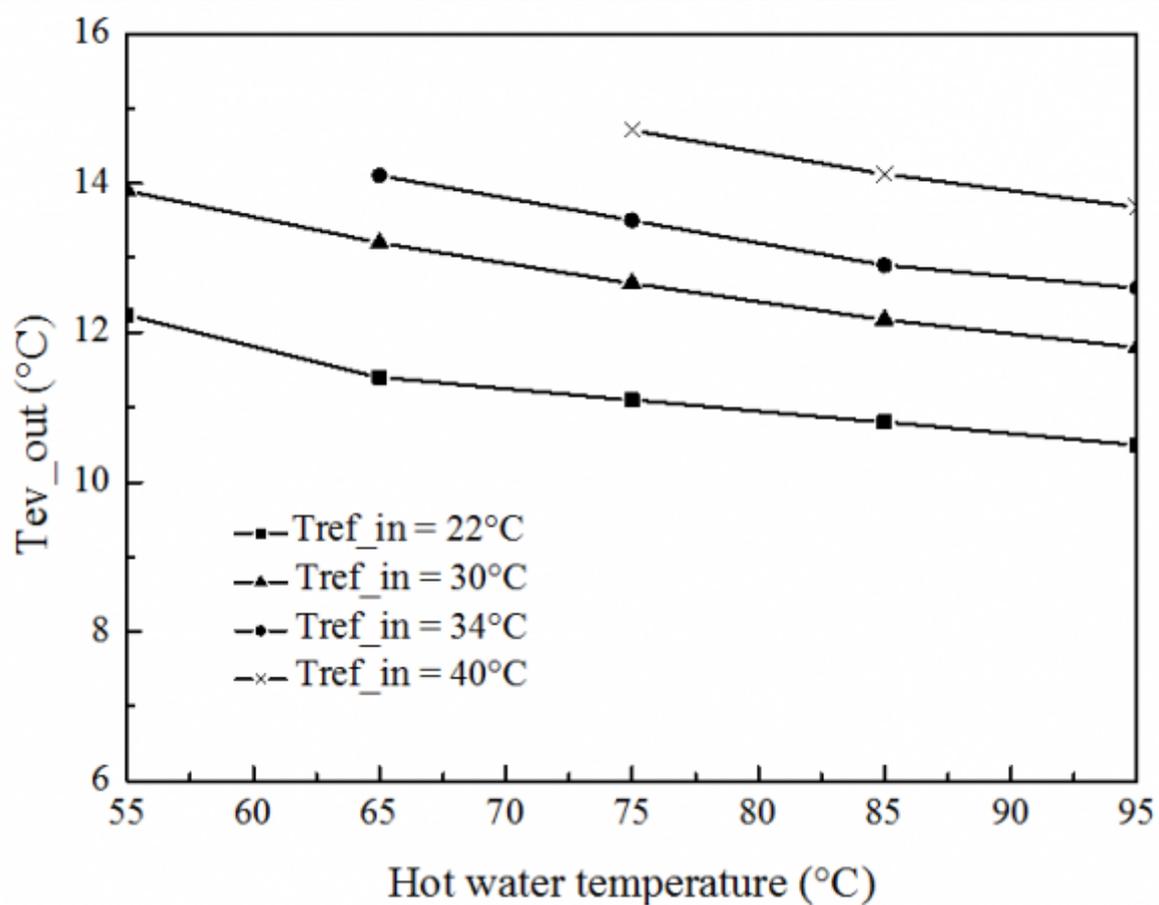
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Figure 3: Fig. 2 :



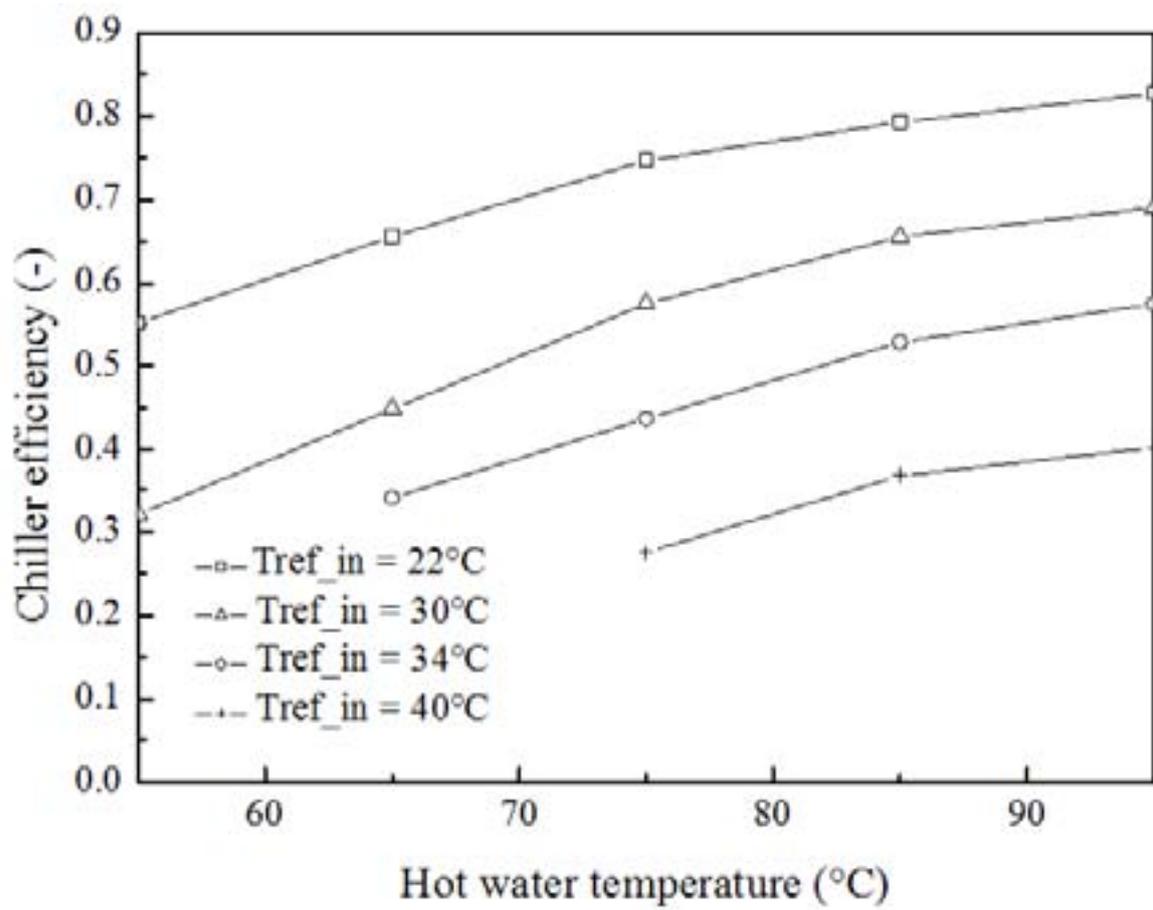
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Figure 4: Fig. 3 :



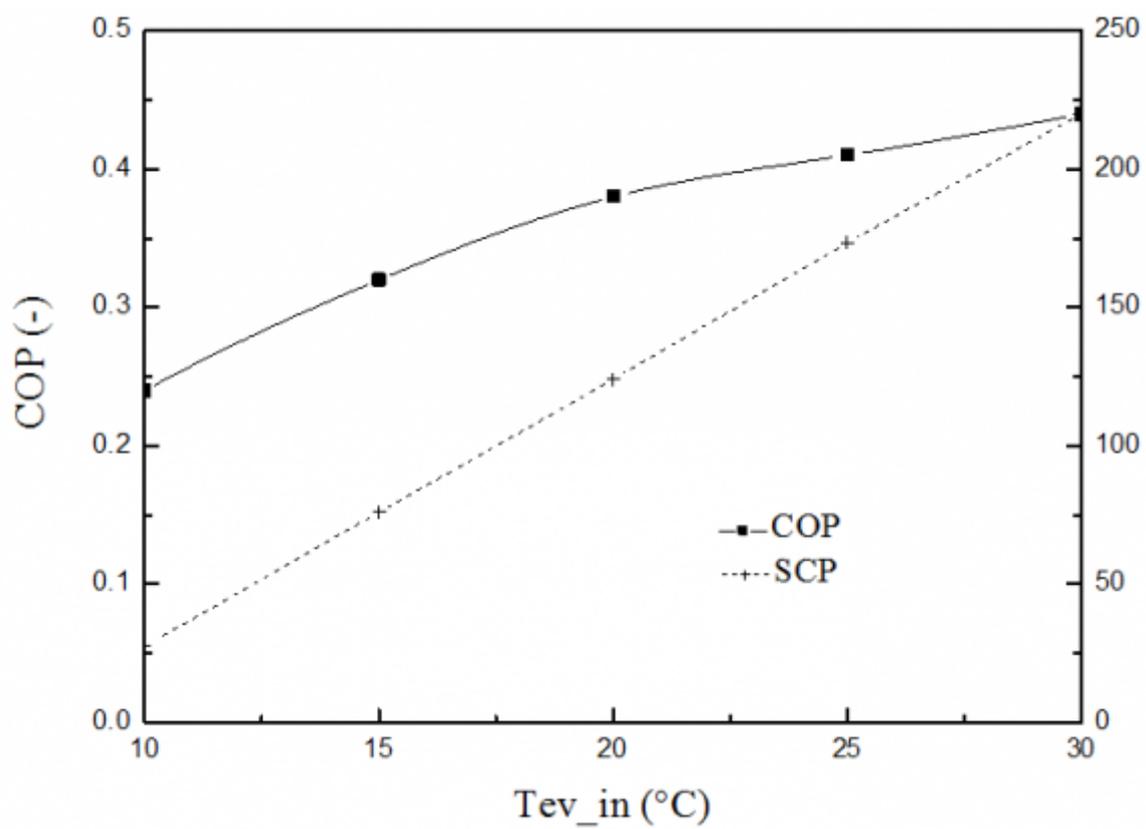
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Figure 5: Fig. 4 :



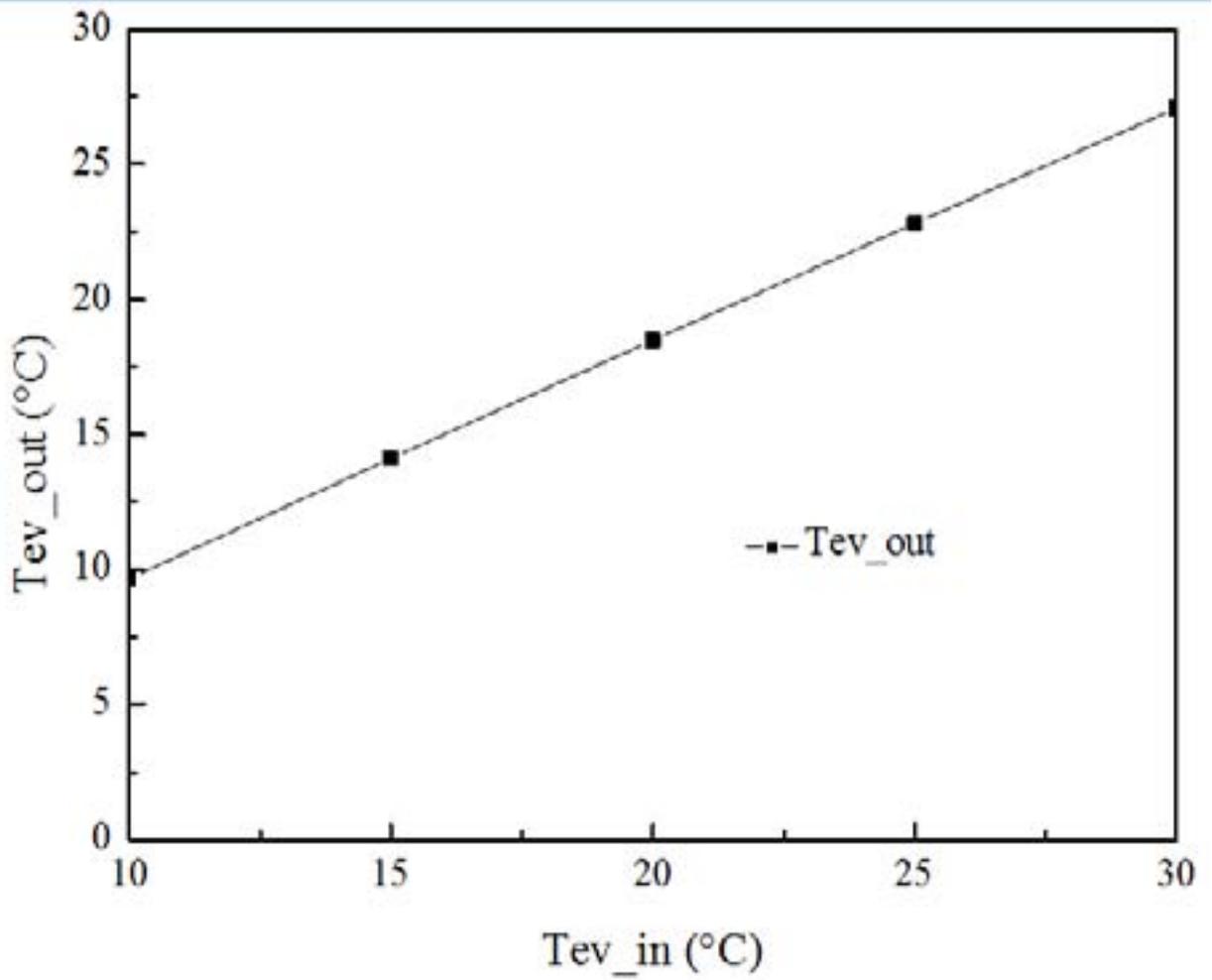
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Figure 6: Fig. 5 :



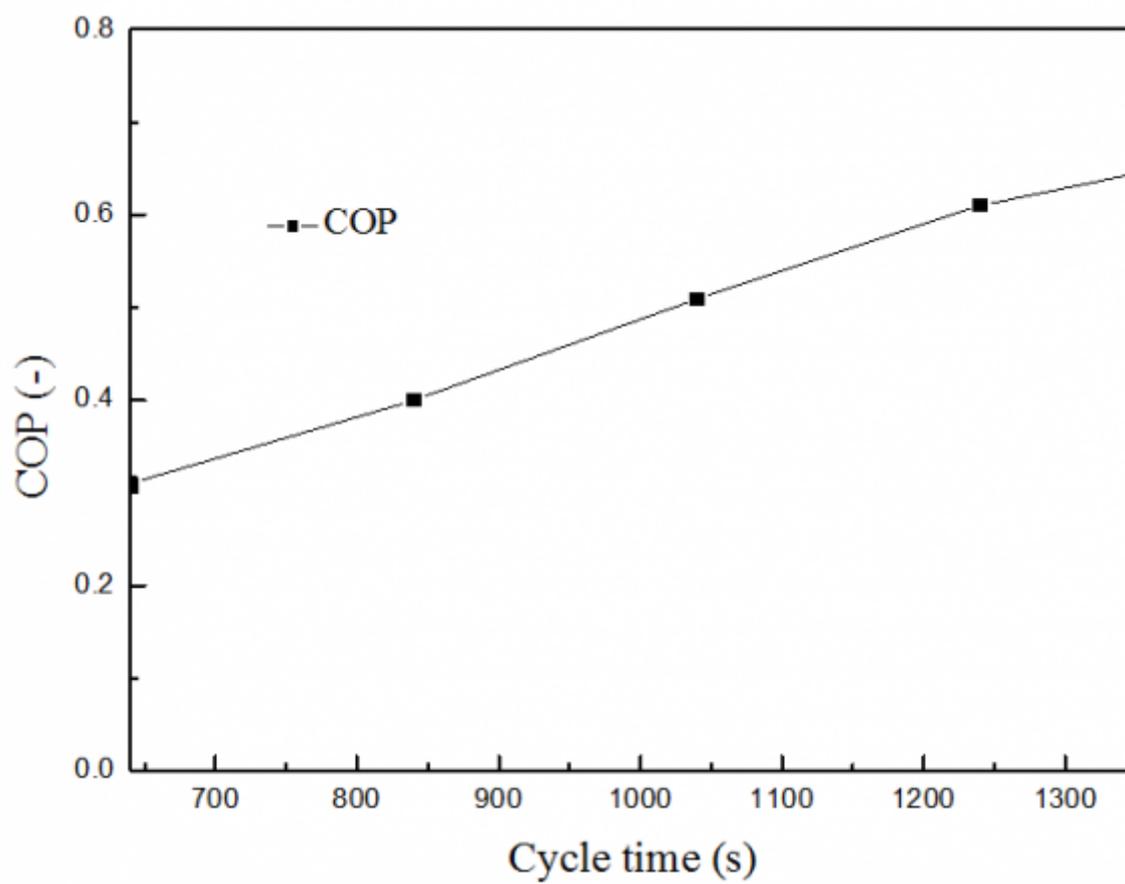
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Figure 7: Fig. 6 :



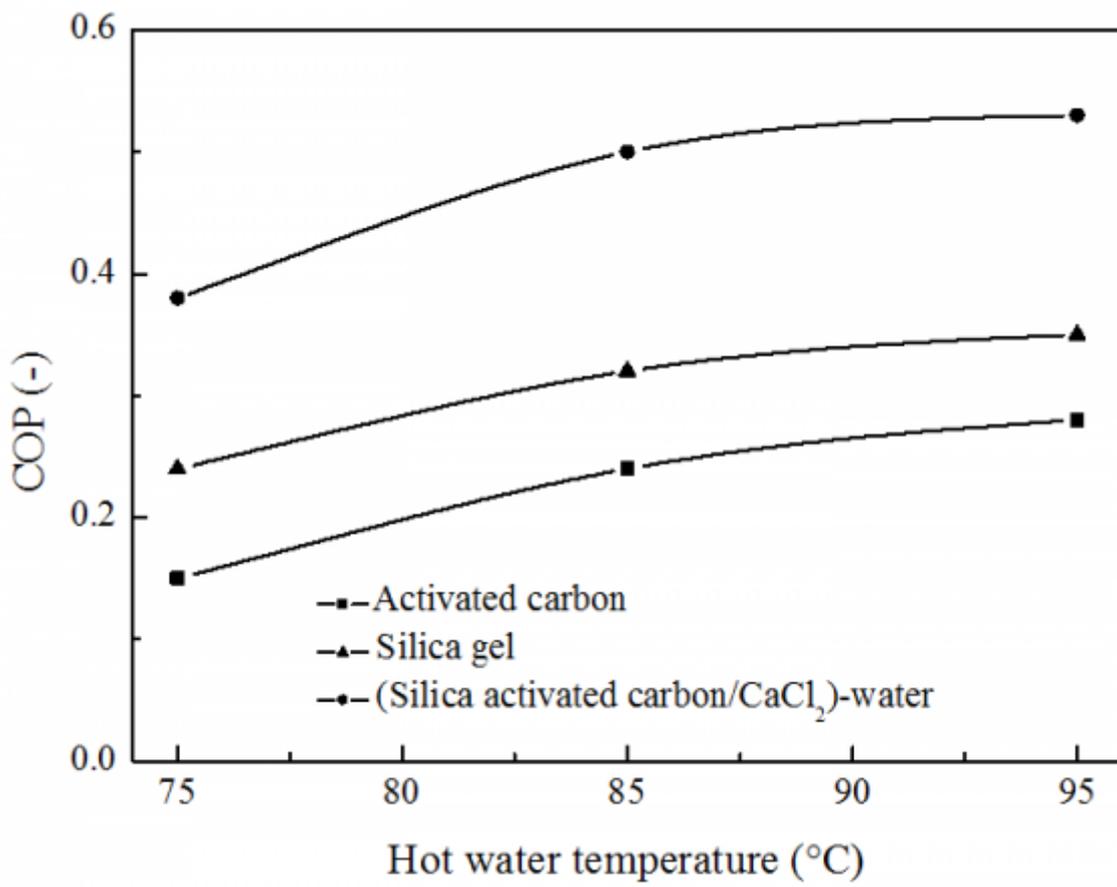
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Figure 8: Figure 7



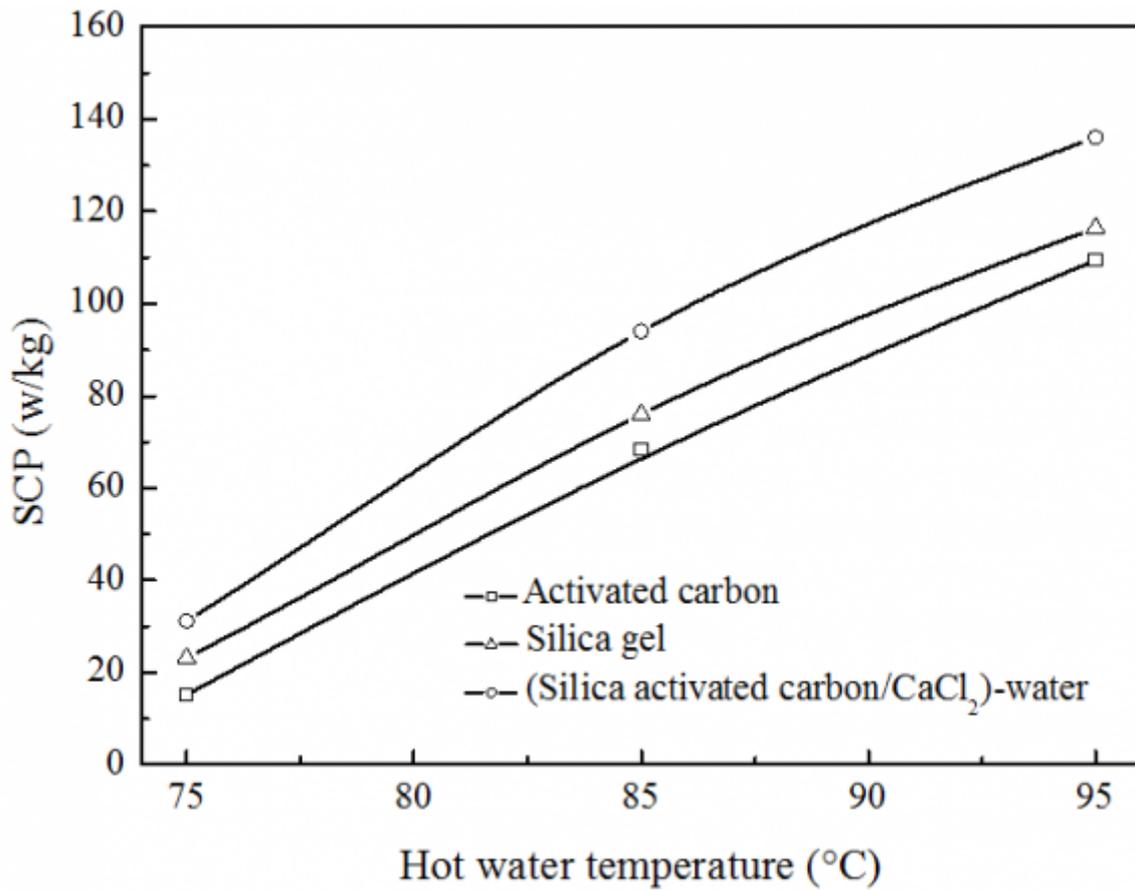
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Figure 9: Fig. 7 :



8

Figure 10: Fig. 8 :



9

Figure 11: Fig. 9 :

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IV. Results and Discussion

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t cycle SCP = m Global Journal t cycle (17) Specific Cooling Power Q ev m a

(18) Where:

[Note: $a = 50 \text{ Kg}$, $C_{p, \text{gel}} = 2800 \text{ kJ/kg}$, $C_{p, \text{water}} = 2500 \text{ kJ/kg}$, $C_{p, \text{CaCl}_2} = 0.386 \text{ kJ/kg.K}$, $C_{p, \text{silica}} = 1.85 \text{ kJ/kg.K}$, $C_{p, \text{activated carbon}} = 0.924 \text{ kJ/kg.K}$, $C_{p, \text{silica activated carbon}} = 4.18 \text{ kJ/kg.k}$, $\dot{m}_{\text{gel}} = 1.6 \text{ m}^3/\text{h}$, $\dot{m}_{\text{water}} = 3.7 \text{ m}^3/\text{h}$, $\dot{m}_{\text{CaCl}_2} = 2 \text{ m}^3/\text{h}$, $T_{\text{gel, in}} = 15^\circ\text{C}$, $T_{\text{gel, out}} = 22^\circ\text{C}$, $T_{\text{water, in}} = 62^\circ\text{C}$ $\dot{m}_{\text{CaCl}_2} = 840 \text{ kg/h}$, a) Model validation Figure]

Figure 12:

1

| | | | |
|--------------------------|---------------------|-------------------------|-------------------------|
| Tde_in | Tref_in | Tev_in | Total cycle time |
| 85 o C | 40 o C | 15 o C | 640-1400 s |
| Pre-heating/cooling time | Hot water flow rate | Cooling water flow rate | Chilled water flow rate |
| 40 s | 1.6 m3/h | 3.7 m3/h | 2 m3/h |

Figure 13: Table 1 :

- 232 [Benelmir. R; Ghilen] , Benelmir. R; Ghilen .
 233 [J. of Thermal Environmental Engineering] , *J. of Thermal & Environmental Engineering* 4 p. .
 234 [Saha et al. ()] , B B Saha , A Akisawa , T Kashiwagi . 2001.
 235 [Ghilen and Descieux ()] , N Ghilen , ; D Descieux . 2014.
 236 [Chihara and Suzuki ()] ‘Air drying by pressure swing adsorption’. K Chihara , M Suzuki . *Journal of Chemical*
 237 *Engineering of Japan* 1983. 16 (4) p. .
 238 [Boelman et al. ()] *Computer simulation of a silica gel-water adsorption refrigeration cycle-the influence of*
 239 *operating conditions of cooling output and COP*, E C Boelman , B B Saha , T Kashiwagi . 1995. ASHRAE
 240 Transactions. p. .
 241 [Núñez et al. ()] ‘Development of an adsorption chiller and heat pump for domestic heating and airconditioning
 242 applications’. Tomas Núñez , , , Walter Mittelbach , Hans-Martin Henning . *Applied Thermal Engineering*
 243 2007. 27 p. .
 244 [He et al. ()] ‘Development of novel type of two-stage adsorption chiller with different adsorbents’. Z H He , H
 245 Y Huang , L S Deng , H R Yuan , N Kobayashi , M Kubota , D D Huhetaoli , Zhao . *The 6th International*
 246 *Conference on Applied Energy -ICAE2014*, 2014.
 247 [Boelman et al. ()] ‘Experimental investigation of a silica gel-water adsorption refrigeration cycle-the influence
 248 of operating conditions on cooling output and COP’. E C Boelman , B B Saha , T Kashiwagi . *ASHRAE*
 249 *Trans* 1995. 101 (2) p. .
 250 [Liu et al. ()] ‘Experimental performance of a silica gel-water adsorption chiller’. Y L Liu , R Z Wang , Z Z Xia
 251 . *Applied Thermal Engineering* 2005. 25 (2-3) p. .
 252 [Liu and Wang ()] ‘Experimental study on a continuous adsorption water chiller with novel design’. Y L Liu , R
 253 Z Wang . *International Journal of Refrigeration* 2005. 28 p. . (Z.Z. Xia.)
 254 [Zaizhong Xia et al. ()] ‘Experimental study on improved two-bed silica gel-water adsorption chiller’. , Zaizhong
 255 Xia , Dechang Wang , Jincui Zhang . *Energy conversion & Management* 2008. p. .
 256 [Sapienza et al. ()] *Influence of the Management Strategy and Operating Conditions on the Performance of an*
 257 *Adsorption Chiller*, A Sapienza , S Santamaria , A Frazzica , A Freni . 2011. Energy. 36 p. .
 258 [Tso et al. ()] ‘Modeling a solar-powered double bed novel composite adsorbent (silica activated carbon/CaCl₂
 259)-water adsorption chiller’. Chi Yan Tso , Sau Chung Fu , Y H Christopher , Chao . BUILD SIMUL 2014.
 260 [Li et al. ()] ‘Performance Analysis of a Multi-Mode Thermochemical Sorption Refrigeration System for Solar-
 261 Powered Cooling’. T X Li , R Z Wang , J K Kiplagat , L Ma . *International Journal of Refrigeration* 2012.
 262 35 p. .
 263 [Tso et al. ()] ‘Performance Analysis of a Waste Heat Driven Activated Carbon Based Composite Adsorbent
 264 -Water Adsorption Chiller Using Simulation Model’. C Y Tso , C Y H Chao , S C Fu . *International Journal*
 265 *of Heat and Mass Transfer* 2012.
 266 [Habib et al. ()] ‘Performance Evaluation of Combined Adsorption Refrigeration Cycles’. K Habib , B B Saha ,
 267 A Chakraborty , S Koyama , K Srinivasan . *International Journal of Refrigeration* 2011. 34 p. .
 268 [Ghilen et al. ()] *Performance of silica gel-water solar adsorption cooling system. Case Studies in Thermal*
 269 *Engineering*, Ghilen , N; Gabsi , S; Messai , S; Benelmir. R; El Ganaoui . 2016. 8 p. .
 270 [Societe et al.] ‘Plateforme Technologique ENERBAT Cogénération -Froid solaire par adsorption -Construction
 271 Bois’. Societe , De Thermique , « Groupe , Thermodynamique . *Journée thématique, Problématiques*
 272 *Scientifiques et Technologiques dans les Procédés Frigorifiques et Thermiques à Sorption*,
 273 [Restuccia et al. ()] ‘Selective water sorbent for solid sorption chiller: experimental results and modeling’. G
 274 Restuccia , A Freni , S Vasta , Yu Aristov . *Int J Refrigeration* 2004. p. .
 275 [Sadeghlu ()] ‘Simulation study of a combined adsorption refrigeration system’. A Sadeghlu . *Applied Thermal*
 276 *Engineering* 2015. 87 p. .
 277 [Solar/waste heat driven two-stage adsorption chiller: the prototype Renewable Energy] ‘Solar/waste heat
 278 driven two-stage adsorption chiller: the prototype’. *Renewable Energy* 23 p. .
 279 [Wang et al. ()] ‘Study of a novel silica gel-water adsorption chiller: part I. Design and performance prediction’.
 280 D C Wang , J Y Wu , Z Z Xia . *Int J Refrig* 2005. 28 (7) p. .
 281 [Chen et al. ()] ‘Study on a compact silica gel-water adsorption chiller without vacuum valves: Design and
 282 experimental study’. C J Chen , R Z Wang , * , Z Z Xia , J K Kiplagat , Z S Lu . *Applied Energy* 2010. 87
 283 p. .
 284 [Habib ()] ‘Study on solar driven combined adsorption refrigeration cycles in tropical climate’. K Habib . *Applied*
 285 *Thermal Engineering* 2013. 50 p. .
 286 [Ganaoui and M; Deuscieux. D; S. Gabsi ()] *Technology Platform ENERBAT -Gas Cogeneration, Solar Heating*
 287 *and Cooling: Int*, El Ganaoui , M; Deuscieux. D; S. Gabsi . 2013.
 288 [Miyazaki et al. ()] ‘The Performance Analysis of a Novel Dual Evaporator Type Three-Bed Adsorption Chiller’.
 289 T Miyazaki , A Akisawa , B B Saha . *International Journal of Refrigeration* 2010. 33 p. .